

PTO-1590 (8-01)

SEARCH REQUEST FORM

Scientific and Technical Information Center

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Requester's Full Name:		Examiner #: /	/ 64) Date: 1 /9/0	<u>ه ک</u>
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If more than one search is subm	iittea, piease prioritiz ********	.e searches in order ***********	or need. *************	******
Please provide a detailed statement of the	search topic: and describe	as specifically as possible	the subject matter to be se	arched.
Include the elected species or structures, k				
utility of the invention. Define any terms	that may have a special me	eaning. Give examples or	relevant citations, authors	, etc, if.
known. Please attach a copy of the cover s	sheet, pertinent claims, and	abstract.	* . *.	
Title of Invention:			• •	
Title of invention.	•		,	
Inventors (please provide full names):			·	
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Earliest Priority Filing Date:		<u>· · ·</u>		
For Sequence Searches Only Please inclu- appropriate serial number.	de all pertinent information (parent, child, divisional, or	issued patent numbers) alon	g with the
appropriate serial number.	•			
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STAFF USE ONLY	Type of Search	Vendors and	l cost where applicable	
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Searcher Prep & Review Time: 2	Fulltext	Sequence Systems		

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FILE 'REGISTRY' ENTERED AT 16:18:58 ON 10 JAN 2002 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2002 American Chemical Society (ACS)

STRUCTURE FILE UPDATES: 8 JAN 2002 HIGHEST RN 381163-99-1 DICTIONARY FILE UPDATES: 8 JAN 2002 HIGHEST RN 381163-99-1

TSCA INFORMATION NOW CURRENT THROUGH July 7, 2001

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Calculated physical property data is now available. See HELP PROPERTIES for more information. See STNote 27, Searching Properties in the CAS Registry File, for complete details: http://www.cas.org/ONLINE/STN/STNOTES/stnotes27.pdf

Juery covers claim 24 510 structures found

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NODE ATTRIBUTES:

NSPEC IS RC AT 6
NSPEC IS RC AT 7
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 7

STEREO ATTRIBUTES: NONE

L5 SCR 1957 AND 1932

L7 510 SEA FILE=REGISTRY SSS FUL L3 AND L5
L8 45 SEA FILE=REGISTRY ABB=ON L7 AND 1-3/TB
L9 4 SEA FILE=REGISTRY ABB=ON L8 AND 3/F

L10 1 SEA FILE=REGISTRY ABB=ON L9 AND C31H44B2F3N12O3STB/MF

=> D L10

L10 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2002 ACS RN 217956-36-0 REGISTRY

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

XU 09/445050 Page 2

- CN Terbium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-(9CI) (CA INDEX NAME)
- MF C31 H44 B2 F3 N12 O3 S Tb
- CI CCS
- SR CA
- LC STN Files: CA, CAPLUS

PAGE 1-A

PAGE 2-A

1 REFERENCES IN FILE CA (1967 TO DATE) 1 REFERENCES IN FILE CAPLUS (1967 TO DATE)

=> FILE HCAPLUS

FILE 'HCAPLUS' ENTERED AT 16:20:16 ON 10 JAN 2002 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2002 AMERICAN CHEMICAL SOCIETY (ACS)

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FILE COVERS 1907 - 10 Jan 2002 VOL 136 ISS 2 FILE LAST UPDATED: 8 Jan 2002 (20020108/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

This file supports REG1stRY for direct browsing and searching of all substance data from the REGISTRY file. Enter HELP FIRST for more information.

HCAplus now provides online access to patents and literature covered in CA from 1907 to the present. Bibliographic information and abstracts were added in 2001 for over 3.8 million records from 1907-1966.

CAS roles have been modified effective December 16, 2001. Please check your SDI profiles to see if they need to be revised. For information on CAS roles, enter HELP ROLES at an arrow prompt or use the CAS Roles thesaurus (/RL field) in this file.

=> S L10

L13 1 L10

=> D BIB L13

L13 ANSWER 1 OF 1 HCAPLUS COPYRIGHT 2002 ACS

AN 1998:806731 HCAPLUS

DN 130:73617

TI Organometallic complexes

IN Christou, Victor

PA Isis Innovation Ltd., UK

SO PCT Int. Appl., 38 pp.

CODEN: PIXXD2

DT Patent

applicant

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XU
        09/445050
                         Page 4
LA
      English
FAN.CNT 1
      PATENT NO.
                              KIND DATE
                                                          APPLICATION NO. DATE
                                                    WO 1998-GB1587 19980601
                                      _____
      WO 9855561 A1 19981210
PΙ
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                            A1 19981221
A1 20000329
                                                        AU 1998-76681
      AU 9876681
                                                                                   19980601
                                                          EP 1998-924488
      EP 988353
                                                                                  19980601
            R: BE, DE, ES, FR, GB, IT, NL
PRAI GB 1997-11237
                             19970602
                                      19980601
      WO 1998-GB1587
OS
      MARPAT 130:73617
RE.CNT 3
RE
(1) Armaroli, N; Chemical Physics Letters 1997, V276(5-6), P435
(2) Univ Princeton; WO 9806242 A 1998 HCAPLUS
(3) Wallac OY; WO 9311433 A 1993 HCAPLUS
     D OUE L16
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L3
                      STR
В 7
                    M 6
NODE ATTRIBUTES:
NSPEC
        IS RC
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NSPEC
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DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS
STEREO ATTRIBUTES: NONE
L5
                      SCR 1957 AND 1932
L7
                510 SEA FILE=REGISTRY SSS FUL L3 AND L5
                  45 SEA FILE=REGISTRY ABB=ON L7 AND 1-3/TB
L8
                   4 SEA FILE=REGISTRY ABB=ON L8 AND 3/F
L9
                   1 SEA FILE=REGISTRY ABB=ON L9 AND C31H44B2F3N12O3STB/MF
L10
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349 SEA FILE=REGISTRY ABB=ON L7 AND 1-3/TB, CE, EU, ER, GD, TM, SM, ND

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09/445050
XU
                        Page 5
             348 SEA FILE=REGISTRY ABB=ON L11 NOT L10
L12
             1 SEA FILE=HCAPLUS ABB=ON L10
76 SEA FILE=HCAPLUS ABB=ON L12
17 SEA FILE=HCAPLUS ABB=ON L14 AND (EL OR ?LUMINESC? OR LIGHT?(4A
L13
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                 ) EMIT?)
              17 SEA FILE=HCAPLUS ABB=ON L13 OR L15
L16
                                             17 CA references from.

Structures + application

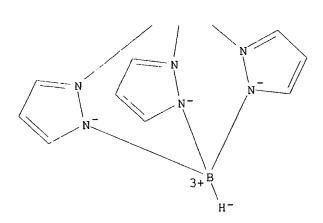
2 ACS includes full CA

rystal structures of two record for

resta complexes: elected Species
=> D L16 ALL 1-17 HITSTR
     ANSWER 1 OF 17 HCAPLUS COPYRIGHT 2002 ACS
L16
     2000:619086 HCAPLUS
ΑN
     133:368714
DN
TI
     Synthesis, luminescence studies and crystal structures of two
     europium(II) hydrotris(pyrazol-1-yl)borate complexes:
     Eu(HBpz3)2.cntdot.2DPSO and Eu(HBpz3)2.cntdot.2BPMU
     Carvalho, A.; Domingos, A.; Isolani, P. C.; Marques, N.; Pires de Matos,
ΑU
     A.; Vicentini, G.
     Departamento de Quimica, ITN, Sacavem, 2686-953, Port.
CS
SO
     Polyhedron (2000), 19(14), 1707-1712
     CODEN: PLYHDE; ISSN: 0277-5387
     Elsevier Science Ltd.
PB
DT
     Journal
LA
     English
CC
     78-7 (Inorganic Chemicals and Reactions)
     Section cross-reference(s): 73, 75
     CASREACT 133:368714
OS
     Europium(II) hydrotris(pyrazol-1-yl)borate complexes, Eu(HBpz3)2L2 (L =
AB
     diphenylsulfoxide, 1; bis(pentamethylene)urea, 2), were synthesized.
     Characterization of 1 and 2 by IR absorption spectroscopy points to
     coordination of the neutral ligands through the sulfoxide and carbonyl
     oxygens, resp. This was corroborated by x-ray diffraction anal. In both
     compds. the metal center is eight-coordinate by way of the six nitrogens
     of the two tridentate pyrazolyl ligands and the oxygen atoms of the
     diphenylsulfoxide or bis(pentamethylene) urea ligands, in a square
     antiprismatic configuration. Both complexes are luminescent at
     77 K and display emission bands with maxima at 610 and 632 nm for 1 and 2,
     resp.
ST
     crystal structure europium hydrotrispyrazolylborato sulfoxide urea;
     europium hydrotrispyrazolylborate sulfoxide urea prepn structure
     luminescence; pyrazolylborate hydrotris europium prepn structure
     luminescence
ΙT
     Molecular structure-property relationship
         (luminescence; of europium(II) hydrotris(pyrazolyl)borato
        diphenylsulfoxide and bis(pentamethylene)urea complexes)
IT
     Crystal structure
       Luminescence
     Molecular structure
         (of europium(II) hydrotris(pyrazolyl)borato diphenylsulfoxide and
        bis(pentamethylene)urea complexes)
     306970-50-3P 306970-52-5P
IT
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
         (prepn., crystal and mol. structure and luminescence)
IT
     945-51-7 5395-04-0, Bis(pentamethylene)urea
     RL: RCT (Reactant)
         (reactant for prepn. of europium(II) hydrotris(pyrazolyl)borato
        complex)
IT
     170126-33-7
     RL: RCT (Reactant)
         (reactant for prepn. of europium(II) hydrotris(pyrazolyl)borato
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diphenylsulfoxide and bis(pentamethylene)urea complexes)
RE.CNT
RE
(1) Alvarez, H; J Coord Chem 1998, V43, P349 HCAPLUS
(2) Bunzli, J; J Alloy Comp 1994, V207, P107
(3) Domingos, A; Polyhedron 1995, V14, P3067 HCAPLUS (4) Drew, M; Coord Chem Rev 1977, V24, P179 HCAPLUS
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(21) Trofimenko, S; J Am Chem Soc 1967, V89, P3170 HCAPLUS
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(24) Zhang, X; New J Chem 1995, V19, P573 HCAPLUS
     306970-50-3P 306970-52-5P
ΙT
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
     (prepn., crystal and mol. structure and luminescence) 306970-50-3 HCAPLUS
RN
     Europium, bis[hydrotris(1H-pyrazolato-.kappa.N1)borato(1-)-
CN
     .kappa.N2',.kappa.N2',.kappa.N2'']bis[1,1'-(sulfinyl-.kappa.O)bis[benzene]]-
     , (SA-8-112222'2'2')- (9CI) (CA INDEX NAME)
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PAGE 1-A

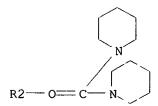


PAGE 2-A

RN 306970-52-5 HCAPLUS

Europium, bis[1,1'-(carbonyl-.kappa.O)bis[piperidine]]bis[hydrotris(1H-pyrazolato-.kappa.N1)borato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-, (SA-8-112222'2'2')- (9CI) (CA INDEX NAME) CN

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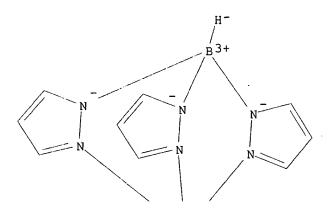


IT 170126-33-7

RL: RCT (Reactant)
 (reactant for prepn. of europium(II) hydrotris(pyrazolyl)borato
 diphenylsulfoxide and bis(pentamethylene)urea complexes)

RN 170126-33-7 HCAPLUS

CN Europium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2'']bis(tetrahydrofuran)- (9CI) (CA INDEX NAME)



PAGE 2-A

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L16
      ANSWER 2 OF 17 HCAPLUS COPYRIGHT 2002 ACS
AN
      2000:227727 HCAPLUS
      132:271477
DN
TΙ
      Divalent lanthanide metal complexes
IN
      Christou, Victor; Salata, Oleg Victorovitch; Shipley, Christopher
PΑ
      Isis Innovation Limited, UK
SO
      PCT Int. Appl., 36 pp.
      CODEN: PIXXD2
DT
      Patent
LA
      English
IC
      ICM C09K011-06
      ICS H05B033-14; C07D231-00
CC
      73-11 (Optical, Electron, and Mass Spectroscopy and Other Related
      Properties)
      Section cross-reference(s): 76, 78
FAN.CNT 1
      PATENT NO.
                            KIND DATE
                                                      APPLICATION NO.
                                                                            DATE
                                                                           19990924
PΙ
      WO 2000018851
                            A1
                                   20000406
                                                      WO 1999-GB3201
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                TJ, TM
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      AU 9961053
                             A1
                                   20000417
                                                      AU 1999-61053
                                                                            19990924
      EP 1115808
                             A1
                                   20010718
                                                      EP 1999-947674
                                                                            19990924
               AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
                IE, SI, LT, LV, FI, RO
PRAI GB 1998-20805
                            Α
                                   19980925
      WO 1999-GB3201
                                   19990924
                             W
AB
      Light-emitting devices are described which employ
      complexes contg. a lanthanide (esp. Eu, Sm or Yb) cation complexed with 1
      to 3 polydentate ligands. The polydentate ligands preferably contain
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.gtoreq.1 lH-pyrazol-1-yl groups, such as tris(lH-pyrazol-1-yl) borate anions. Selected complexes are claimed, as are methods for prepg. them by reacting the divalent cation with the complex ions in soln. and then sepg. the product from the soln.

ST divalent lanthanide polydentate ligand complex electroluminescent device

IT Electroluminescent devices

(divalent lanthanide metal complexes with polydentate ligands and their prepn. and **electroluminescent** devices using them)

IT Phosphors

(electroluminescent; divalent lanthanide metal complexes with polydentate ligands and their prepn. and electroluminescent devices using them)

IT 2085-33-8, Tris(8-hydroxyquinolinato)aluminum 65181-78-4, N,N'-Diphenyl-N,N'-bis(3-methylphenyl)-1,1'-biphenyl-4,4'-diamine 150405-69-9, 3-(4-Biphenylyl)-4-phenyl-5-(4-tert-butylphenyl)-1,2,4-triazole

RL: DEV (Device component use); USES (Uses)

(divalent lanthanide metal complexes with polydentate ligands and their prepn. and **electroluminescent** devices using them)

IT 151305-99-6P 151306-01-3P 171672-52-9P 263570-02-1P 263570-55-4P 263570-62-3P 263571-07-9P

RL: DEV (Device component use); IMF (Industrial manufacture); PRP (Properties); PREP (Preparation); USES (Uses)

(divalent lanthanide metal complexes with polydentate ligands and their prepn. and **electroluminescent** devices using them)

IT 10010-93-2P 121314-30-5P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)

(divalent lanthanide metal complexes with polydentate ligands and their prepn. and **electroluminescent** devices using them)

IT 367-57-7 7803-57-8, Hydrazine monohydrate 13762-51-1, Potassium borohydride 13874-77-6, Ytterbium dichloride 17567-17-8 84768-84-3 94138-28-0 123882-26-8 157409-94-4 RL: RCT (Reactant)

(divalent lanthanide metal complexes with polydentate ligands and their prepn. and **electroluminescent** devices using them)

RE CNT 6

RE

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- (5) Victor, C; WO 9855561 A 1998 HCAPLUS
- (6) Zhang, X; J CHEM 1995, V19(5-6), P573 HCAPLUS
- IT 151305-99-6P 171672-52-9P 263570-02-1P 263570-55-4P 263570-62-3P 263571-07-9P

RL: DEV (Device component use); IMF (Industrial manufacture); PRP (Properties); PREP (Preparation); USES (Uses)

(divalent lanthanide metal complexes with polydentate ligands and their prepn. and electroluminescent devices using them)

RN 151305-99-6 HCAPLUS

CN Samarium, bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-, (OC-6-1'1')- (9CI) (CA INDEX NAME)

PAGE 2-A

RN 171672-52-9 HCAPLUS

CN Europium, bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-, (OC-6-1'1')- (9CI) (CA INDEX NAME)

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PAGE 2-A

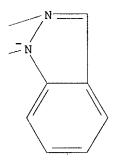
RN 263570-02-1 HCAPLUS

Europium, bis[hydrotris(1H-indazolato-.kappa.N1)borato(1-).kappa.N2,.kappa.N2',.kappa.N2'']-, (OC-6-1'1')- (9CI) (CA INDEX NAME) CN

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PAGE 1-A B3+

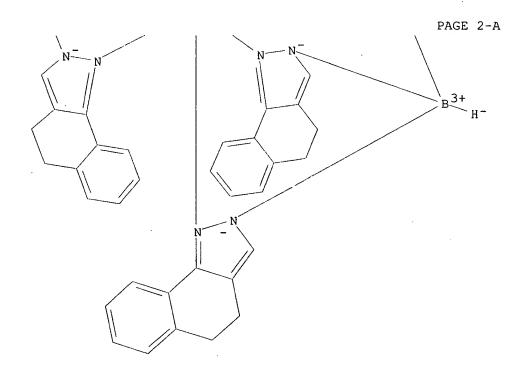
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RN 263570-55-4 HCAPLUS

Europium, bis[tris(4,5-dihydro-2H-benz[g]indazolato-.kappa.N2)hydroborato(1-)-.kappa.N1,:kappa.N1',.kappa.N1'']-, (OC-6-1'1')-CN (9CI) (CA INDEX NAME)

PAGE 1-A



RN 263570-62-3 HCAPLUS CN

Europium, bis[hydrotris(lH-pyrazolato-.kappa.N1)borato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-, (OC-6-1'1')- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 2-A

RN

263571-07-9 HCAPLUS Europium, bis[hydrotris[5-methyl-3-(trifluoromethyl)-1H-pyrazolato-.kappa.N1]borato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-, (OC-6-1'1')-CN (9CI) (CA INDEX NAME)

F3C N CF3 N Me Me Me H-

PAGE 2-A

L16 ANSWER 3 OF 17 HCAPLUS COPYRIGHT 2002 ACS

AN 1999:720372 HCAPLUS

DN 132:72779

TI Structural and Photophysical Properties of Mononuclear and Dinuclear Lanthanide(III) Complexes of Multidentate Podand Ligands Based on Poly(pyrazoly1)borates

AU Armaroli, Nicola; Accorsi, Gianluca; Barigelletti, Francesco; Couchman, Samantha M.; Fleming, James S.; Harden, Nicholas C.; Jeffery, John C.; Mann, Karen L. V.; McCleverty, Jon A.; Rees, Leigh H.; Starling, Sarah R.; Ward, Michael D.

- CS Istituto di Fotochimica e Radiazioni d'Alta Energia del CNR, Bologna, 40129, Italy
- SO Inorg. Chem. (1999), 38(25), 5769-5776 CODEN: INOCAJ; ISSN: 0020-1669
- PB American Chemical Society
- DT Journal
- LA English
- CC 78-7 (Inorganic Chemicals and Reactions) Section cross-reference(s): 73, 75
- Lanthanide(III) complexes were prepd. with [L1] [the tetradentate AB chelating ligand bis{3-(2-pyridyl)pyrazolyl}dihydroborate], [L2]- [the tetradentate chelating ligand bis{3-(2-pyrazinyl)pyrazolyl}dihydroborate], [L3] - [the hexadentate chelating ligand bis[3-{6'-(2,2'-bipyridyl)}pyrazol-1-yl]dihydroborate], and [L4]2- [the 12-dentate compartmental ligand hexakis(3-(2-pyridyl)pyrazol-1-yl}diboran(IV)ate, which has two hexadentate tris(pyrazolyl)borate-based cavities linked back-to-back by a B-B bond]. [Ln(L1)2(NO3)] are 10-coordinate with two tetradentate N-donor ligands and one bidentate nitrate. [Ln(L2)2(NO3)] have 10-coordinate structures similar to those of the [L1]- complexes except that the coordinated N1 of the pyrazine rings is not such a good donor as the pyridine rings in the [L1]- complexes, leading to marked lengthening of these Ln-N bonds. [Ln(L3)(NO3)2] are also 10-coordinate from one hexadentate chelating ligand which has a pseudoequatorial coordination mode and two pseudoaxial bidentate nitrate ligands; the hexadentate ligand has a shallow helical twist to prevent steric interference between its ends. Finally $[{Ln(NO3)2}2(L4)]$ are dinuclear, with each metal center being 10-coordinate from a tripodal hexadentate ligand cavity and two bidentate nitrates. Five complexes were structurally characterized: [Tb(L2)2(NO3)].cntdot.DMF is monoclinic (space group P21/c) with a 14.881(3), b 15.5199(12), c 15.845(2) .ANG., .beta. 92.387(12).degree., and Z = 4. [Gd(L2)2(NO3)].cntdot.DMF is monoclinic (space group P21/c) with a 14.926(2), b 15.465(2), c 15.878(2) .ANG., .beta. 92.698(11).degree., and Z = 4. [Eu(L3)(NO3)2].cntdot.DMF.cntdot.0.5Et20 is triclinic (P.hivin.1) with a 10.020(3), b 13.036(3), c 14.740(3) .ANG., .alpha. 70.114(14), .beta. 71.55(2), .gamma. 79.66(2).degree., and Z = 2. [{La(NO3)(DMF)2}2(L4)](NO3)2.cntdot.DMF is orthorhombic (Pbca) with a 18.813(2), b 15.241(2), c 27.322(2), and Z = 4. $[\{Gd(NO3)2\}2(L4)].cntdot.2.4DMF$ is tetragonal (P42/n) with a 16.622(6), c 24.19(5) .ANG., and Z = 4. Detailed photophys. studies were performed on the free ligands and their complexes with Gd(III), Eu(III), and Tb(III) in several solvents. The results show a wide range in the emission properties of the complexes which can be rationalized in terms of subtle variations in the steric and electronic properties of the ligands. In particular the dinuclear Tb(III) complex of [L4]2- has an emission quantum yield of .apprx.0.5 in D2O and MeOD.
- ST crystal structure lanthanide polypyrazolylborate multidentate podand nitrato; lanthanide polypyrazolylborate multidentate podand nitrato prepn; photophys property lanthanide polypyrazolylborate multidentate podand nitrato; luminescence lanthanide polypyrazolylborate multidentate podand nitrato
- IT Crystal structure

Luminescence

Molecular structure

(of lanthanide poly(pyrazolyl)borate-based multidentate podand nitrato complexes)

IT Podands

RL: PRP (Properties); RCT (Reactant)
 (photophys. properties of poly(pyrazolyl)borate-based multidentate
 podands and their complexes with lanthanide(III))

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Rare earth complexes
IT
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (poly(pyrazolyl)borate; prepn., structure, and photophys. properties of
        lanthanide poly(pyrazolyl)borate-based multidentate podand nitrato
        complexes)
IT
     188606-26-0
                    192885-45-3
                                  212397-25-6
                                                 226876-49-9
     RL: PRP (Properties); RCT (Reactant)
        (complexation with lanthanide(III) and photophys. properties of)
IT
     253150-94-6P 253151-01-8P 253151-07-4P
     253151-15-4P 253151-23-4P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and crystal structure of)
IT
     192885-29-3P 205534-50-5P 253150-66-2P
     253150-68-4P 253150-69-5P 253150-73-1P
     253150-75-3P 253150-85-5P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and photophys. properties of)
IT
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     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
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ΙT
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     253150-72-0P
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RE.CNT
RE
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- 253150-94-6P 253151-01-8P 253151-07-4P

253151-23-4P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and crystal structure of)

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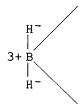
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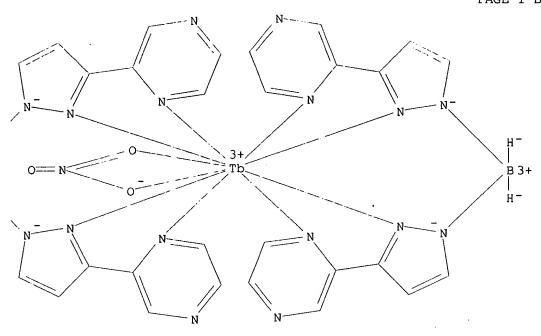
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CCI CCS



ΧU

PAGE 1-B



CM 2

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RN

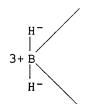
253151-01-8 HCAPLUS
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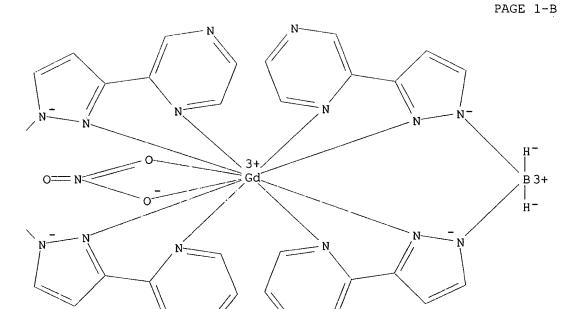
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CMF C28 H24 B2 Gd N17 O3

CCI CCS





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XU 09/445050 Page 23

RN 253151-07-4 HCAPLUS

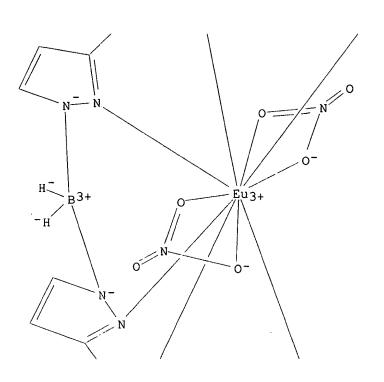
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CM 1

CRN 253150-72-0 CMF C26 H20 B Eu N10 O6

CCI CCS





PAGE 2-A



PAGE 3-A

CM 2

CRN 68-12-2 CMF C3 H7 N O

СНЗ н3С-й-Сн = О

> CM 3

CRN 60-29-7 CMF C4 H10 O ${\rm H_{3}C-CH_{2}-O-CH_{2}-CH_{3}}$

RN 253151-23-4 HCAPLUS

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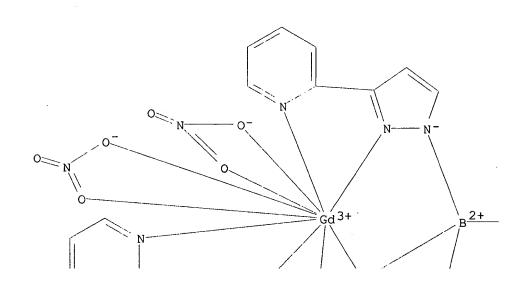
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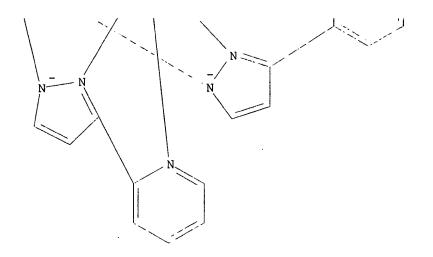
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PAGE 2-B



2 CM

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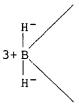
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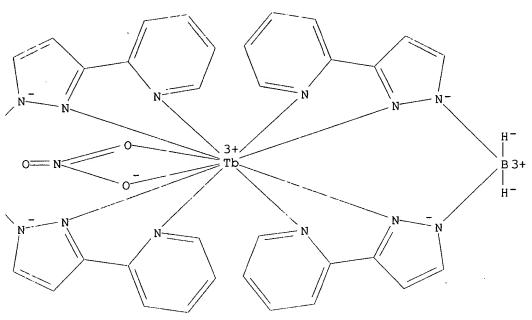
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RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and photophys. properties of)

192885-29-3 HCAPLUS RN

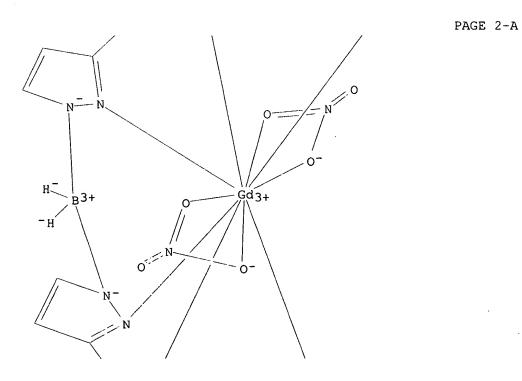
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RN 205534-50-5 HCAPLUS
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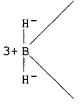


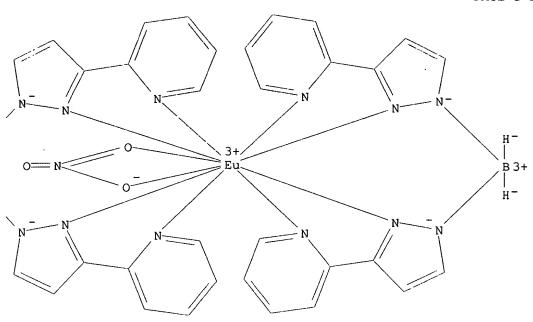
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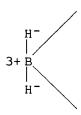
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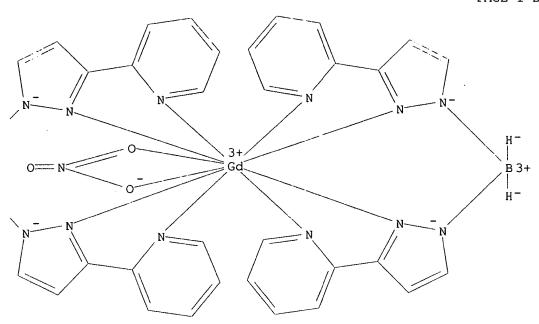




RN

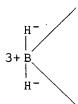
Gadolinium, bis[dihydrobis[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-)](nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME) CN

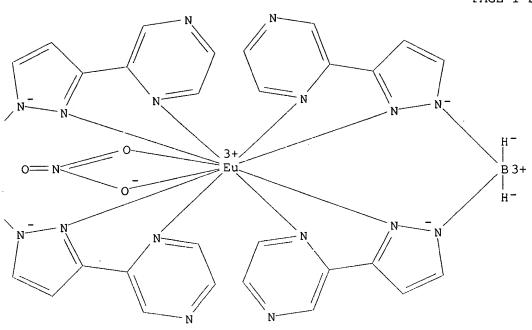




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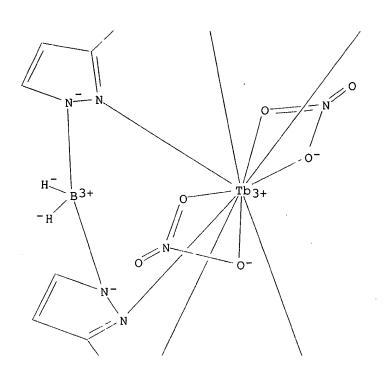
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RN 253150-73-1 HCAPLUS Terbium, [dihydrobis[6-(1H-pyrazol-3-yl-.kappa.N1)-2,2'-bipyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME) CN





PAGE 2-A



PAGE 3-A

RN

253150-75-3 HCAPLUS
Europium, [.mu.-[hexakis[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]diborato(2-)]]tetrakis(nitrato-.kappa.O,.kappa.O')di- (9CI) (CA INDEX NAME) CN

PAGE 1-B

RN 253150-85-5 HCAPLUS
CN Terbium, [.mu.-[hexakis[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]diborato(2)]]tetrakis(nitrato-.kappa.O,.kappa.O')di- (9CI) (CA INDEX NAME)

PAGE 1-B

212397-22-3P 253150-70-8P 253150-71-9P IT 253150-72-0P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., mol. structure, and photophys. properties of) 212397-22-3 HCAPLUS

RN

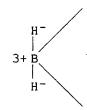
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Gadolinium, [.mu.-[hexakis[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]diborato(2-)]]tetrakis(nitrato-.kappa.O,.kappa.O')di-(9CI) (CA INDEX NAME)

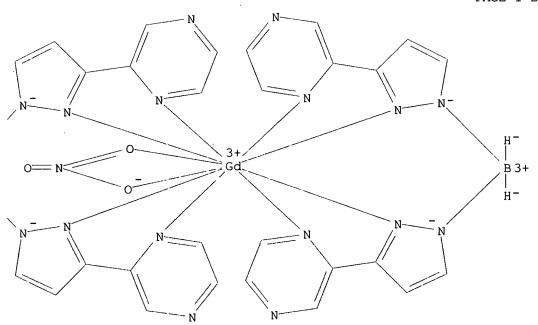
PAGE 1-B

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CN Gadolinium, bis[dihydrobis[(1H-pyrazol-3-yl-.kappa.N1)pyrazinato]borato(1)](nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)

PAGE 1-A

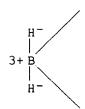


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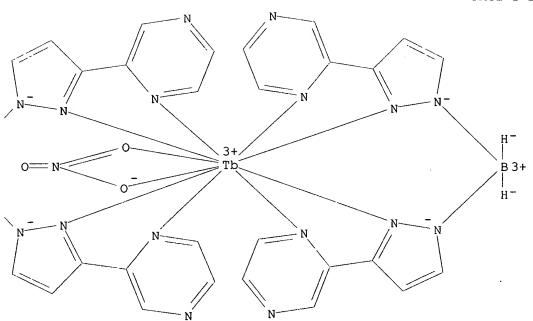


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253150-71-9 HCAPLUS
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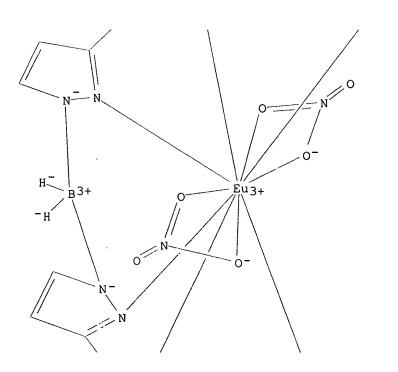
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RN 253150-72-0 HCAPLUS

Europium, [dihydrobis[6-(1H-pyrazol-3-yl-.kappa.N1)-2,2'-bipyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX CN NAME)





PAGE 3-A



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L16
    ANSWER 4 OF 17 HCAPLUS COPYRIGHT 2002 ACS
AN
     1999:327093 HCAPLUS
     131:25525
DN
ΤI
     Orange electroluminescence from a divalent europium complex
     Shipley, Christopher P.; Capecchi, Simone; Salata, Oleg V.; Etchells,
ΑU
     Mark; Dobson, Peter J.; Christou, Victor
CS
     Inorganic Chemistry Laboratory, Department Chemistry, University Oxford,
     Oxford, OX1 3QR, UK
SO
     Adv. Mater. (Weinheim, Ger.) (1999), 11(7), 533-536
     CODEN: ADVMEW; ISSN: 0935-9648
PB
     Wiley-VCH Verlag GmbH
     Journal
DT
LA
     English
     73-12 (Optical, Electron, and Mass Spectroscopy and Other Related
     Properties)
     Section cross-reference(s): 76, 78
    A thin-film electroluminescent (EL) device based on a
AB
     divalent mol. Eu complex was prepd. by CVD of luminescent Tp2Eu
     (bis(tris(dimethylpyrazolyl)borate)europium(II)) as org. layer on
    ITO-coated glass. The Tp2Eu complex exhibited bright orange photoluminescence (PL) with a high PL quantum efficiency in the
     solid state. The volatility of the compd. enabled the prepn. of
    multilayer EL devices with a sandwich structure, in which the
     org. layer was embedded between an electron transporting layer and a hole
     transporting layer. Thin film EL devices with this structure
     showed visible emission comparable to the PL spectrum of the pure Eu
     complex.
ST
     europium methylpyrazolylborate complex prepn electroluminescent
     device fabrication
TΤ
    Electroluminescent devices
        (fabrication with europium tris(dimethylpyrazolyl)borate complex)
TΤ
    Luminescence
       Luminescence, electroluminescence
     UV and visible spectra
        (of europium tris(dimethylpyrazolyl)borate complex)
IT
     2085-33-8, Tris(8-hydroxyquinolinato)aluminum
                                                      65181-78-4, TPD
     150405-69-9, TAZ
     RL: DEV (Device component use); USES (Uses)
        (electroluminescent device fabrication with europium
        methylpyrazolylborate complex and)
IT
     171672-52-9P
     RL: DEV (Device component use); PRP (Properties); SPN (Synthetic
     preparation); PREP (Preparation); USES (Uses)
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fabrication with)

RL: RCT (Reactant)

17567-17-8

(prepn. and UV absorption and luminescence and electroluminescence and electroluminescent device

(reaction with europium iodo THF complex in prepn. of europium tris(dimethylpyrazolyl)borate complex) RE.CNT RE (1) Antoniadis, H; Polym Adv Technol 1998, V8, P392 (2) Berardini, M; Inorg Chem 1995, V34, P6179 HCAPLUS (3) Chen, C; Macromol Symp 1998, V125, P1 HCAPLUS (4) Edwards, A; Synth Met 1997, V84, P433 HCAPLUS (5) Harkonen, G; J All Comp 1995, V225, P552 (6) Jiang, J; Coord Chem Rev 1998, V170, P1 HCAPLUS (7) Jones, P; Inorg Chem 1997, V36, P10 HCAPLUS (8) Kido, J; J All Comp 1993, V192, P30 HCAPLUS (9) Kido, J; Jpn J Appl Phys Pt 2 1993, V32, P917 (10) Kido, J; Jpn J Appl Phys Pt 2 1996, V35, P394 (11) Kina, H; Jpn J Appl Phys Pt 1 1997, V36, P150 HCAPLUS (12) King, C; J Vac Sci 1996, V14, P1729 HCAPLUS (13) Kraft, A; Angew Chem Int Ed 1998, V37, P402 (14) Leskela, M; Mater Chem Phys 1987, V16, P349 HCAPLUS (15) Li, W; Synth Met 1997, V91, P263 HCAPLUS (16) Liu, L; Synth Met 1997, V91, P267 HCAPLUS (17) Liu, S; Inorg Chem 1996, V35, P76 HCAPLUS (18) Ma, D; Chin J Chem 1998, V16, P1 HCAPLUS (19) Moratti, S; Synth Met 1995, V71, P2117 HCAPLUS (20) Rothberg, L; J Mater Res 1996, V11, P3174 HCAPLUS (21) Sabbatini, N; Coord Chem Rev 1993, V123, P201 HCAPLUS (22) Santos, I; New J Chem 1995, V19, P551 HCAPLUS (23) Sato, H; Polym Adv Technol 1997, V8, P454 HCAPLUS (24) Sheats, J; Science 1996, V273, P884 HCAPLUS (25) Staring, E; Synth Met 1995, V71, P2179 HCAPLUS (26) Takats, J; Organometallics 1993, V12, P4286 HCAPLUS (27) Trofimenko, S; Chem Rev 1993, V93, P943 HCAPLUS (28) Wang, P; Adv Mater 1996, V8, P237 HCAPLUS (29) Zhang, X; Appl Phys Lett 1997, V71, P2596 HCAPLUS ΙT 171672-52-9P RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses) (prepn. and UV absorption and luminescence and electroluminescence and electroluminescent device fabrication with) RN171672-52-9 HCAPLUS Europium, bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.Nl)hydroborato(1-)-CN

.kappa.N2,.kappa.N2',.kappa.N2'']-, (OC-6-1'1')- (9CI) (CA INDEX NAME)

PAGE 1-A

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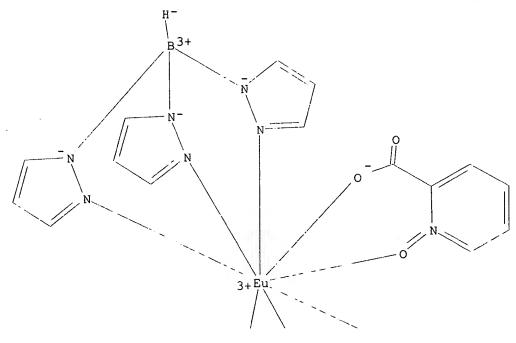
PAGE 2-A

- L16 ANSWER 5 OF 17 HCAPLUS COPYRIGHT 2002 ACS
- AN 1999:141492 HCAPLUS
- DN 130:261027
- TI Heteroleptic poly(pyrazol-1-yl)borate derivatives of the lanthanides. The syntheses of picolinate-N-oxide complexes and the x-ray crystal structure of [Tb{HB(C3H3N2)3}2(ONC5H4CO2-2)]
- AU Lawrence, Royston G.; Jones, Christopher J.; Kresinski, Roman A.
- CS School of Chemistry, University of Birmingham, Birmingham, B15 2TT, UK
- SO Inorg. Chim. Acta (1999), 285(2), 283-289

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CODEN: ICHAA3; ISSN: 0020-1693
PB
   Elsevier Science S.A.
DT
     Journal
LA
     English
CC
     78-7 (Inorganic Chemicals and Reactions)
     Section cross-reference(s): 73, 75 [Ln(Tp)2(pnx)] {Ln = Y, Eu, Gd, Tb, Er, Yb or Lu; Tp-=
AB
     hydrotris(pyrazol-1-yl)borate, pnxH = picolinic acid N-oxide) were
     synthesized and characterized. The complexes which contain Eu or Tb are
     emissive upon ligand excitation and substantially larger intensities of
     emission are assocd. with excitation of the pnx- ligand than with
     excitation of the Tp- ligand. The solid state structure (monoclinic,
     space group P21/a, R1 = 4.21%) of [Tb(Tp)2(pnx)] shows the complex to be
     eight-coordinate and monomeric, contg. a pnx- co-ligand chelating via one
     N-oxide oxygen and one carboxylate oxygen. The geometry around the Tb3+
     ion is distorted square antiprismatic with a steric angle sum of 0.79.
ST
     crystal structure terbium hydridotrispyrazolylborato picolinate oxide
     complex; terbium hydridotrispyrazolylborato picolinate oxide complex prepn
     structure luminescence; rare earth hydridotrispyrazolylborate
     picolinate oxide complex prepn structure luminescence; borate
     hydridotrispyrazolyl rare earth picolinate oxide prepn structure
     luminescence; pyrazolylborate hydridotris rare earth picolinate
     oxide prepn structure luminescence
ΙT
     Luminescence
        (of rare earth hydridotris(pyrazolyl)borato picolinate oxide
        heteroleptic complexes)
IT
     Crystal structure
     Molecular structure
        (of terbium hydridotris(pyrazolyl)borato picolinate oxide heteroleptic
        complex)
IT
     Rare earth amine complexes
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (poly(pyrazolyl)borate; prepn., crystal structure and
        luminescence of rare earth hydridotris(pyrazolyl)borato
        picolinate oxide heteroleptic complexes)
IT
     824-40-8, Picolinic acid N-oxide
                                         18583-60-3, Potassium
     hydrotris(pyrazol-1-yl)borate
     RL: RCT (Reactant)
        (for prepn. of rare earth hydridotris(pyrazolyl)borato picolinate oxide
        heteroleptic complexes)
IT
     221457-36-9P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and luminescence)
     221457-35-8P 221457-37-0P 221457-39-2P
                                               221457-40-5P
IT
     221457-41-6P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (prepn. of)
IT
     221457-38-1P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn., crystal structure and luminescence)
       31
RE.CNT
RE
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- ΙT 221457-36-9P
 - RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and luminescence)
- 221457-36-9 HCAPLUS RN
- CN Europium, bis[hydrotris(1H-pyrazolato-.kappa.N1)borato(1-)-.kappa.N2,.kappa.N2',.kappa.N2''][(2-pyridinecarboxylic acid-.kappa.O2) 1-(oxidato-.kappa.O)]- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 2-A B3+ н-

221457-37-0P 221457-39-2P ΙT

RL: SPN (Synthetic preparation); PREP (Preparation)

RN

(prepn. of)

221457-37-0 HCAPLUS

Gadolinium, bis[hydrotris(1H-pyrazolato-.kappa.N1)borato(1-).kappa.N2,.kappa.N2',.kappa.N2''][(2-pyridinecarboxylic acid-.kappa.O2)
1-(oxidato-.kappa.O)]- (9CI) (CA INDEX NAME) CN

PAGE 1-A 3+dq

PAGE 2-A ₿3+ Н –

RN

221457-39-2 HCAPLUS Erbium, bis[hydrotris(1H-pyrazolato-.kappa.N1)borato(1-)-.kappa.N2,.kappa.N2',.kappa.N2''][(2-pyridinecarboxylic acid-.kappa.O2)1-(oxidato-.kappa.O)]- (9CI) (CA INDEX NAME) CN

PAGE 1-A

PAGE 2-A ₿3+ н-

IT 221457-38-1P

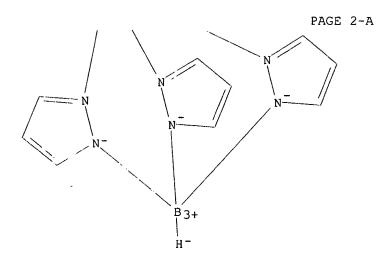
> RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., crystal structure and luminescence) 221457-38-1 HCAPLUS

RN

CN Terbium, bis[hydrotris(1H-pyrazolato-.kappa.N1)borato(1-)-.kappa.N2',.kappa.N2''][(2-pyridinecarboxylic acid-.kappa.O2)
1-(oxidato-.kappa.O)]-, (SA-8-123333'3')- (9CI) (CA INDEX NAME)

PAGE 1-A

ΧU



ANSWER 6 OF 17 HCAPLUS COPYRIGHT 2002 ACS

1999:55807 HCAPLUS AN

DN 130:245582

- TΙ Lanthanide complexes of a new sterically hindered potentially hexadentate podand ligand based on a tris(pyrazolyl)borate core; crystal structures, solution structures and luminescence properties
- Reeves, Zoe R.; Mann, Karen L. V.; Jeffery, John C.; McCleverty, Jon A.; ΑU Ward, Michael D.; Barigelletti, Francesco; Armaroli, Nicola
- School of Chemistry, University of Bristol, Bristol, BS8 1TS, UK J. Chem. Soc., Dalton Trans. (1999), (3), 349-356 CS
- SO CODEN: JCDTBI; ISSN: 0300-9246
- PB Royal Society of Chemistry
- DTJournal
- LA English
- 78-7 (Inorganic Chemicals and Reactions) CC Section cross-reference(s): 28, 73, 75
- The new podand ligand hydrotris[3-(6-methyl)pyridin-2-ylpyrazol-1-AΒ yl]borate [L1] - was prepd. which contains three bidentate pyrazolyl/pyridine arms attached to a {BH}- head-group. This ligand differs from an earlier ligand hydrotris[3-(2-pyridyl)pyrazol-1-yl]borate [L2]- by the presence of Me groups attached to the C6 positions of the pyridyl rings, which would interfere with each other sterically if the ligand coordinated in a fully hexadentate manner. Instead, crystallog. anal. of [M(L1)(NO3)2(H2O)] (M = Eu, Tb or Gd) showed that partial dissocn. of the podand occurs to relieve this potential steric problem: either one or two of the pyridyl groups are not coordinated, such that [L1] - is penta- or tetra-dentate, but instead are involved in intramol. N.cntdot..cntdot..cntdot.H-O hydrogen-bonding interactions with the coordinated water mol. The presence of both structural forms in single crystals of the gadolinium and europium complexes shows that interconversion between them in soln. must be facile. Variable-temp. 1H NMR spectra of the diamagnetic lanthanum(III) analog shows that, whereas all three ligand arms are equiv. on the NMR timescale at high temps., at -80.degree. there is mirror symmetry in the complex such that two arms are equiv. and the 3rd is different from the other two; this is consistent with the cryst. form in which [L1]- is tetradentate with two pendant pyridyl arms, which has pseudo-mirror symmetry. Luminescence studies showed that whereas the ligand-based luminescence is retained in the gadolinium(III) complex, in the europium(III) and

ST

IT

ΙT

ΙT

09/445050 Page 53 terbium(III) complexes the ligand-centered emission is quenched by ligand-to-metal energy transfer, resulting in the usual metal-centered emission spectra. The intensity of the emission from the europium(III) and terbium(III) complexes of [L1] - is substantially reduced compared to the emission from the analogous complexes [M(L2)(NO3)2] (M = Eu or Tb) which the authors ascribe to the sterically induced poorer coordination of the podand ligand, resulting in (i) less efficient ligand-to-metal energy transfer, and (ii) coordination of labile solvent mols. (H2O) to the metal centers. rare earth hydrotrispyridinylpyrazolylborate complex prepn crystal structure luminescence; lanthanide hydrotrispyridinylpyrazolylbo rate complex prepn crystal structure luminescence; europium hydrotrispyridinylpyrazolylborate complex prepn crystal structure luminescence; terbium hydrotrispyridinylpyrazolylborate complex prepn crystal structure luminescence; gadolinium hydrotrispyridinylpyrazolylborate complex prepn crystal structure luminescence; pyridinylpyrazolylborate rare earth complex prepn crystal structure luminescence; pyrazolylborate rare earth complex prepn crystal structure luminescence Intramolecular energy transfer Luminescence quenching Photoinduced energy transfer (luminescence quenching in europium and terbium $\ensuremath{\mathbf{C}}$ hydrotris(methylpyridinylpyrazolyl)borate complexes due to ligand-to-metal energy transfer) Crystal structure Luminescence Molecular structure (of rare earth hydrotris(methylpyridinylpyrazolyl)borate complexes) Rare earth complexes RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., crystal structure and luminescence of lanthanide hydrotris (methylpyridinylpyrazolyl)borate complexes) 6940-57-4P, 2-Acetyl-6-methylpyridine 79571-43-0P 203569-23-7P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)

IT (intermediate in prepn. of hydrotris(methylpyridinylpyrazolyl)borate and its rare earth complexes)

ΙT 221367-97-1P

> RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation) (prepn. and complexation with rare earths)

IT 221367-80-2P

> RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and crystal structure and luminescence)

IT 221367-73-3P

RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of)

221367-50-6P 221367-60-8P ΙT

> RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., crystal structure and luminescence)

IT 221367-65-3P

> RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., mol. structure and luminescence)

IT 4637-24-5, Dimethylformamide dimethyl acetal 5315-25-3, 2-Bromo-6-methylpyridine 13762-51-1, Potassium tetrahydroborate

RL: RCT (Reactant)

(reactant for prepn. of hydrotris(methylpyridinylpyrazolyl)borate and its rare earth complexes)

RE.CNT 28

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CCI

CCS

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CM 2

CRN 75-09-2

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

ΧU 09/445050

CMF C H2 C12

C1-CH2-C1

IT 221367-50-6P 221367-60-8P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., crystal structure and luminescence) 221367-50-6 HCAPLUS

RN

Europium, aqua[hydrotris[2-methyl-6-(1H-pyrazol-3-yl-CN

Page 56

.kappa.N1)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')-, compd.

with aqua[hydrotris[2-methyl-6-(1H-pyrazol-3-yl-

.kappa.N1)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')europium (1:1) (9CI) (CA INDEX NAME)

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CRN 221367-49-3

CMF C27 H27 B Eu N11 O7

CCI CCS

PAGE 1-A

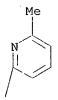
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CMF C27 H27 B Eu N11 O7

CCI CCS

PAGE 1-A



RN

221367-60-8 HCAPLUS
Gadolinium, aqua[hydrotris[2-methyl-6-(1H-pyrazol-3-yl-.kappa.Nl)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')-, compd. with aqua[hydrotris[2-methyl-6-(1H-pyrazol-3-yl-.kappa.Nl)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')gadolinium CN

(1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 221367-59-5

C27 H27 B Gd N11 O7 CMF

CCS CCI

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$$O = N$$

$$N = N$$

CM 2

221367-58-4

CMF C27 H27 B Gd N11 O7

CCI CCS

TI221367-65-3P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., mol. structure and luminescence)

RN 221367-65-3 HCAPLUS

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

09/445050 Page 61 ΧU

Terbium, aqua[hydrotris[2-methyl-6-(lH-pyrazol-3-yl.kappa.Nl)pyridinato]borato(l-)]bis(nitrato-.kappa.O,.kappa.O')- (9CI)
(CA INDEX NAME) CN

PAGE 1-A

PAGE 2-A

$$-H = \frac{3+}{100} = \frac{3+}{100} = \frac{100}{100} = \frac{100}{100}$$

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L16
     ANSWER 7 OF 17 HCAPLUS COPYRIGHT 2002 ACS
AN
     1999:9912 HCAPLUS
     130:102684
DN
ΤI
     Electroluminescent material
IN
     Kathirgamanathan, Poopathy
     South Bank University Enterprises Ltd., UK
PA
SO
     PCT Int. Appl., 39 pp.
     CODEN: PIXXD2
DT
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LA
     English
IC
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     73-11 (Optical, Electron, and Mass Spectroscopy and Other Related
CC
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PΙ
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             FI, FR,
             CM, GA, GN, ML, MR, NE, SN, TD, TG
     AU 9881165
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     EP 990016
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PRAI GB 1997-12483
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     WO 1998-GB1773
OS
     MARPAT 130:102684
AB
     Electroluminescent devices comprising a transparent substrate on
     which is formed a layer of an electroluminescent material are
     described in which the electroluminescent material is a rare
     earth metal, actinide or transition metal org. complex which has a
     photoluminescent efficiency (PL) >25%, preferably >40%.
     Electroluminescent complexes are also described. in which the
     metal is a rare earth, transition metal, lanthanide, or an actinide and
     .gtoreq.1 of the ligands is either O-C(R')-C(R'')-C(R'')-O or a
     2,2'-Bis(pyridyl)ketone deriv. (R'= (un)substituted arom. or heterocylic
     ring structures, a hydrocarbyl of a fluorocarbon, or tert-butyl; and R" =
     (un) substituted arom. or heterocylic ring structures, a hydrocarbyl of a
     fluorocarbon, F, or H, or can be part of a copolymer). Preferably, the
     metals are selected from Sm(III), Eu(III), Tb(III), Dy(III), Yb(III),
     Lu(III), Gd (III), Eu(II), U(III), UO2(VI), and Th(III).
ST
     electroluminescent device metal complex; material
     electroluminescent metal complex
IT
     Actinide compounds
     RL: DEV (Device component use); USES (Uses)
        (complexes; electroluminescent materials based on metal
        complexes and devices using them)
ΙT
     Electroluminescent devices
       Electroluminescent phosphors
        (electroluminescent materials based on metal complexes and
        devices using them)
IT
     Rare earth complexes
```

Transition metal complexes RL: DEV (Device component use); USES (Uses) (electroluminescent materials based on metal complexes and devices using them) ΙT Polyanilines RL: DEV (Device component use); USES (Uses) (hole transport material; electroluminescent materials based on metal complexes and devices using them) IT 7429-90-5, Aluminium, uses 7439-93-2, Lithium, uses 7439-95-4, 7440-70-2, Calcium, uses Magnesium, uses 37271-44-6 RL: DEV (Device component use); USES (Uses) (anode; electroluminescent materials based on metal complexes and devices using them) IT 50926-11-9, Indium tin oxide RL: DEV (Device component use); USES (Uses) (cathode; electroluminescent materials based on metal complexes and devices using them) 1118-71-4D, terbium-dipyrazolyl oxide borate and terbium-tripyrazolyl ΙT oxide borate complexes 7439-94-3D, Lutetium, complexes 7440-19-9D, 7440-27-9D, Terbium, dipivaloylmethane-dipyrazolyl Samarium, complexes oxide borate and dipivaloylmethane-tripyrazolyl oxide borate complexes 7440-54-2D, Gadolinium, complexes 7440-64-4D, Ytterbium, complexes 7440-29-1D, Thorium, complexes 7440-61-1D, Uranium, complexes 219121-79-6D, terbium dipivaloylmethane complexes 20219-51-6 219121-80-9D, terbium dipivaloylmethane complexes 219136-83-1 219136-94-4 219136-89-7 219136-85-3 219136-98-8 219137-01-6 219137-06-1 RL: DEV (Device component use); USES (Uses) (electroluminescent materials based on metal complexes and devices using them) 156952-11-3P IT 156915-57-0P 156952-13-5P 203806-96-6P 219121-71-8P 219121-73-0P 219121-74-1P 219121-72-9P 219121-75-2P 219121-76-3P 219121-78-5P RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses) (electroluminescent materials based on metal complexes and devices using them) 541-09-3, Uranyl acetate 1662-01-7, 4,7-Diphenyl-1,10-phenanthroline TΤ 14552-07-9 15522-69-7 19437-26-4, Di-(2-pyridyl) ketone 2156-69-6 31239-06-2, Imidotetraphenyldiphosphinic acid 218917-64-7 218917-67-0 218917-70-5 219144-50-0 RL: RCT (Reactant) (electroluminescent materials based on metal complexes and devices using them) IT 15492-51-0P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation) (electroluminescent materials based on metal complexes and devices using them) 2085-33-8, Tris(8-hydroxyquinolinato)aluminum ΙT RL: DEV (Device component use); USES (Uses) (electron-injecting material; electroluminescent materials based on metal complexes and devices using them) 25233-30-1, Polyaniline IT 25067-59-8, Poly(vinylcarbazole) N, N'-Diphenyl-N, N'-bis(3-methylphenyl)-1, 1'-biphenyl-4, 4'-diamine RL: DEV (Device component use); USES (Uses) (hole transport material; electroluminescent materials based on metal complexes and devices using them) RE.CNT (1) Amersham Int Plc; EP 0556005 A 1993 HCAPLUS

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- (8) Lin, L; International Conference on Electroluminescence of Molecular Materials and Related Phenomena 1997, V91(1-3), P267
- IT 20219-51-6 219136-83-1

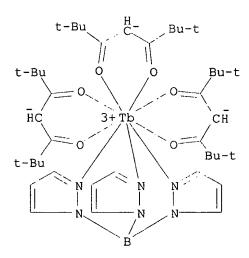
RL: DEV (Device component use); USES (Uses)

(${f electroluminescent}$ materials based on metal complexes and devices using them)

RN 20219-51-6 HCAPLUS

CN Terbium, tris(2,2,6,6-tetramethyl-3,5-heptanedionato-

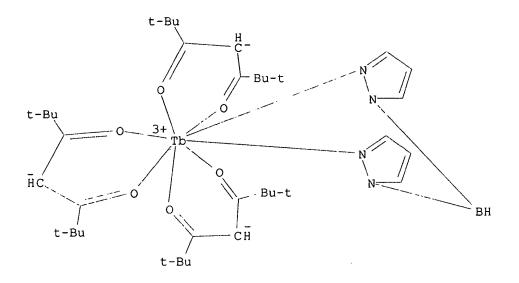
.kappa.O,.kappa.O')[1,1',1''-borylidynetris[1H-pyrazole-.kappa.N2]]- (9CI)
 (CA INDEX NAME)



RN 219136-83-1 HCAPLUS

CN Terbium, [1,1'-borylenebis[1H-pyrazole-.kappa.N2]]tris(2,2,6,6-tetramethyl-3,5-heptanedionato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)

1.16



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ANSWER 8 OF 17 HCAPLUS COPYRIGHT 2002 ACS
AN
       1998:806731 HCAPLUS
                                                                    applicant only reference to only reference to
DN
       130:73617
TΙ
       Organometallic complexes
IN
       Christou, Victor
PΑ
       Isis Innovation Ltd., UK
SO
       PCT Int. Appl., 38 pp.
       CODEN: PIXXD2
DT
       Patent
LA
       English
IC
       ICM C09K011-06
       ICS H05B033-14; C07D401-00
       73-5 (Optical, Electron, and Mass Spectroscopy and Other Related
CC
       Properties)
       Section cross-reference(s): 74, 76, 78
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                              KIND
                                       DATE
                                                            APPLICATION NO.
                                                                                    DATE
       WO 9855561
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                                       19981210
                                                            WO 1998-GB1587
                                                                                    19980601
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                                                                                    19980601
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                                                                                    19980601
            R: BE, DE, ES, FR, GB, IT, NL
PRAI GB 1997-11237
                                       19970602
       WO 1998-GB1587
                                       19980601
os
       MARPAT 130:73617
AΒ
       Light-emitting devices are described which employ
       organometallic complexes comprising a lanthanide metal cation complexed
```

with 1-3 polydentate ligands contg. .gtoreq.1 (un)substituted pyrazolyl groups optionally fused with (un)substituted heterocyclic or carbocyclic ST

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(non)arom. ring systems, with a coordinate bond formed between the metal and one of the nitrogen atoms of the pyrazolyl rings. Preferably, the ligands comprise trispyrazolyl borate derivs. Organometallic compds. suitable for the devices are also claimed, as are methods of producing them entailing the reaction of the ligands with a cation followed by sepn. of the products. Compns. combining the compds. with a matrix material are also described. Use in electroluminescent flat panel displays is also described. lanthanide pyrazolyl deriv complex electroluminescent material Electroluminescent devices Electroluminescent phosphors (lanthanide-pyrazolyl deriv. complexes and electroluminescent devices and displays using them) Rare earth complexes RL: DEV (Device component use); USES (Uses) (lanthanide-pyrazolyl deriv. complexes and electroluminescent devices and displays using them) 15082-28-7 25067-59-8, Polyvinylcarbazole RL: DEV (Device component use); USES (Uses) (lanthanide-pyrazolyl deriv. complexes and electroluminescent devices and displays using them) 171672-48-3P 171672-50-7P 171672-51-8P 217956-36-0P 217956-37-1P 217956-38-2P 217956-39-3P 217956-40-6P 217956-41-7P 217956-42-8P 217956-43-9P 217956-44-0P 217956-45-1P 217956-46-2P 217956-47-3P 217956-48-4P 217956-49-5P 217956-50-8P 217956-51-9P 217956-52-0P 217956-53-1P RL: DEV (Device component use); IMF (Industrial manufacture); PRP (Properties); PREP (Preparation); USES (Uses) (lanthanide-pyrazolyl deriv. complexes and electroluminescent devices and displays using them) 10025-76-0, Europium trichloride 13762-51-1, Potassium borohydride 14704-41-7 17567-17-8 34622-08-7, Neodymium triflate 52093-25-1, Europium triflate 52093-28-4, Samarium triflate 52093-29-5, Gadolinium triflate 76089-77-5, Cerium(III) trifluoromethylsulfonate 84768-84-3 139177-64-3, Erbium triflate 141478-68-4, Thulium triflate 148980-31-8, Terbium(III) trifluoromethylsulfonate 157409-94-4 217956-54-2 RL: RCT (Reactant) (lanthanide-pyrazolyl deriv. complexes and electroluminescent devices and displays using them) 167898-36-4P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation) (lanthanide-pyrazolyl deriv. complexes and electroluminescent devices and displays using them) RE.CNT (1) Armaroli, N; Chemical Physics Letters 1997, V276(5-6), P435 (2) Univ Princeton; WO 9806242 A 1998 HCAPLUS (3) Wallac OY; WO 9311433 A 1993 HCAPLUS 171672-48-3P 171672-50-7P 171672-51-8P 217956-36-0P 217956-37-1P 217956-38-2P 217956-39-3P 217956-40-6P 217956-41-7P 217956-42-8P 217956-43-9P 217956-44-0P 217956-45-1P 217956-46-2P 217956-47-3P 217956-48-4P 217956-49-5P 217956-50-8P 217956-51-9P 217956-52-0P 217956-53-1P RL: DEV (Device component use); IMF (Industrial manufacture); PRP (Properties); PREP (Preparation); USES (Uses)

(lanthanide-pyrazolyl deriv. complexes and electroluminescent devices and displays using them)

171672-48-3 HCAPLUS RN

Cerium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-, (OCF-7-1-2222'2')- (9CI) (CA INDEX NAME) CN

PAGE 1-A

PAGE 2-A

PAGE 3-A

RN 171672-50-7 HCAPLUS

CN Neodymium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-, (OCF-7-1-2222'2')- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 3-A

RN 171672-51-8 HCAPLUS

CN Samarium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-, (OCF-7-1-2222'2'2')- (9CI) (CA INDEX NAME)

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PAGE 2-A

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RN

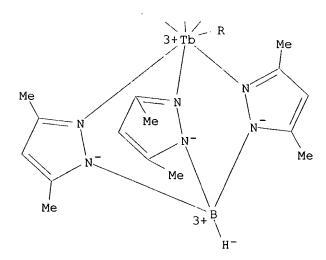
217956-36-0 HCAPLUS

CN Terbium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-(9CI) (CA INDEX NAME)

PAGE 1-A

et elected

PAGE 2-A



217956-37-1 HCAPLUS RN

Europium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-(9CI) (CA INDEX NAME) CN

PAGE 1-A

PAGE 2-A

PAGE 3-A

RN 217956-38-2 HCAPLUS Erbium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-(9CI) (CA INDEX NAME) CN

PAGE 2-A

217956-39-3 HCAPLUS RN

CN Europium, chlorobis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']- (9CI) (CA INDEX NAME)

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PAGE 2-A

RN 217956-40-6 HCAPLUS CN

Europium, chlorobis[tris[3,5-bis(trifluoromethyl)-1H-pyrazolato-.kappa.N1]hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']- (9CI) (CA INDEX NAME)

PAGE 1-A

PAGE 2-A

RN 217956-41-7 HCAPLUS

CN Thulium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']-(9CI) (CA INDEX NAME)

PAGE 2-A

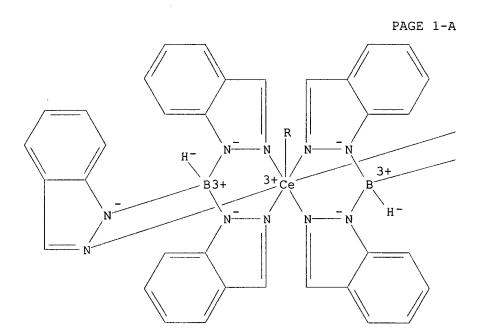
RN 217956-42-8 HCAPLUS

Gadolinium, (trifluoromethanesulfonato-.kappa.O)bis[tris(3,5-dimethyl-1H-pyrazolato-.kappa.N1)hydroborato(1-)-.kappa.N2,.kappa.N2',.kappa.N2'']- (9CI) (CA INDEX NAME) CN

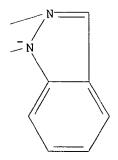
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PAGE 2-A

PAGE 3-A

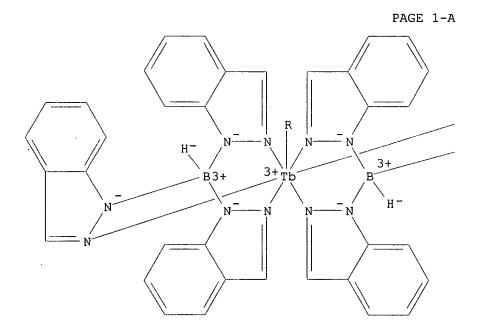


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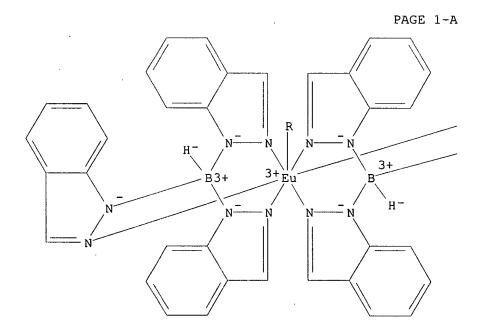
PAGE 2-A

RN 217956-44-0 HCAPLUS
CN Terbium, bis[hydrotris(1H-indazolato-.kappa.N1)borato(1-).kappa.N2,.kappa.N2',.kappa.N2''](trifluoromethanesulfonato-.kappa.O)(9CI) (CA INDEX NAME)



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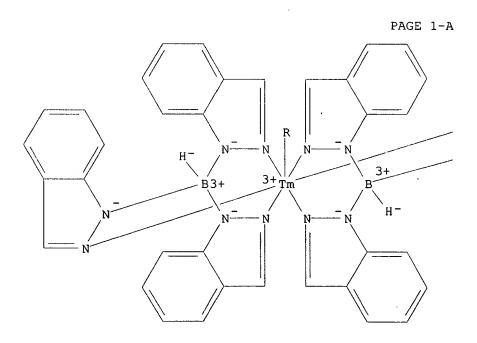
PAGE 2-A



PAGE 2-A

RN 217956-46-2 HCAPLUS

CN Thulium, bis[hydrotris(lH-indazolato-.kappa.N1)borato(l-).kappa.N2,.kappa.N2',.kappa.N2''](trifluoromethanesulfonato-.kappa.O)(9CI) (CA INDEX NAME)



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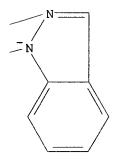
PAGE 2-A

RN

217956-47-3 HCAPLUS
Samarium, bis[hydrotris(1H-indazolato-.kappa.N1)borato(1-).kappa.N2,.kappa.N2',.kappa.N2''](trifluoromethanesulfonato-.kappa.O)(9CI) (CA INDEX NAME) CN

PAGE 1-A 3+ 3+\sm в́3+

PAGE 1-B



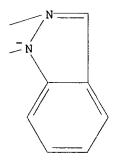
PAGE 2-A

RN

217956-48-4 HCAPLUS
Gadolinium, bis[hydrotris(1H-indazolato-.kappa.N1)borato(1-).kappa.N2,.kappa.N2',.kappa.N2''](trifluoromethanesulfonato-.kappa.O)-CN (9CI) (CA INDEX NAME)

PAGE 1-A **B**3+ N-

PAGE 1-B



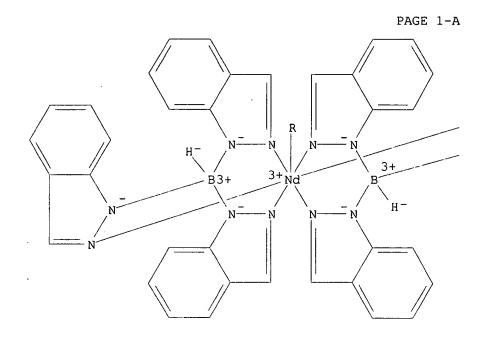
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RN 217956-49-5 HCAPLUS

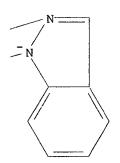
Neodymium, bis[hydrotris(1H-indazolato-.kappa.N1)borato(1-)-.kappa.N2,.kappa.N2',.kappa.N2''](trifluoromethanesulfonato-.kappa.O)-CN

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(9CI) (CA INDEX NAME)



PAGE 1-B



PAGE 2-A

217956-50-8 HCAPLUS RN

Terbium, (trifluoromethanesulfonato-.kappa.O)bis[tris(4,5-dihydro-2H-CN

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

benz[g]indazolato-.kappa.N2)hydroborato(1-)-.kappa.N1,.kappa.N1',.kappa.N1
'']- (9CI) (CA INDEX NAME)

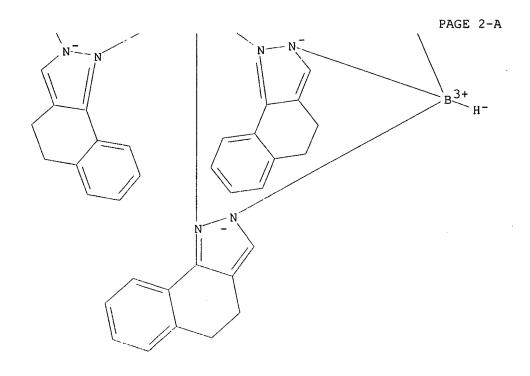
PAGE 1-A

PAGE 3-A

RN 217956-51-9 HCAPLUS

CN Cerium, (trifluoromethanesulfonato-.kappa.O)bis[tris(4,5-dihydro-2H-benz[g]indazolato-.kappa.N2)hydroborato(1-)-.kappa.N1,.kappa.N1',.kappa.N1']- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 3-A

RN 217956-52-0 HCAPLUS

CN Europium, (trifluoromethanesulfonato-.kappa.O)bis[tris(4,5-dihydro-2H-benz[g]indazolato-.kappa.N2)hydroborato(1-)-.kappa.N1,.kappa.N1',.kappa.N1'']- (9CI) (CA INDEX NAME)

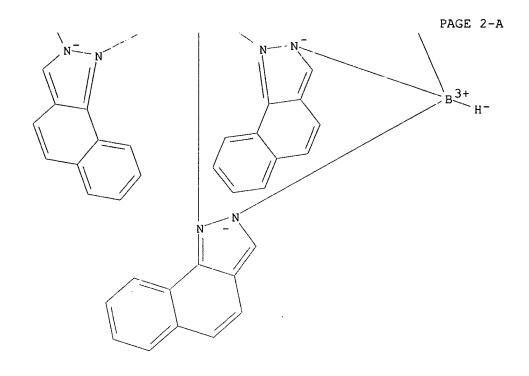
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PAGE 3-A

RN 217956-53-1 HCAPLUS

CN Terbium, (trifluoromethanesulfonato-.kappa.O)bis[tris(2H-benz[g]indazolato-.kappa.N2)hydroborato(1-)-.kappa.N1,.kappa.N1',.kappa.N1'']- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 3-A

L16 ANSWER 9 OF 17 HCAPLUS COPYRIGHT 2002 ACS

1998:577455 HCAPLUS AN

DN 129:239129

TΙ Heterodinuclear Complexes Containing d- and f-block Elements: Synthesis, Structural Characterization, and Metal-Metal Interactions of Novel Chromium(III) - Lanthanide(III) Compounds Bridged by Oxalate

Sanada, Takayuki; Suzuki, Takayoshi; Yoshida, Takafumi; Kaizaki, Sumio ΑU

Department of Chemistry Graduate School of Science, Osaka University, CS Toyonaka, 560, Japan

Inorg. Chem. (1998), 37(18), 4712-4717
CODEN: INOCAJ; ISSN: 0020-1669 SO

PB American Chemical Society

 DT Journal

LA English

CC 78-7 (Inorganic Chemicals and Reactions) Section cross-reference(s): 73, 75, 77

The reaction of Ln(III) ions with a tripodal ligand HBpz3-AΒ (hydrotris(pyrazol-1-yl)borate) and a complex ligand [Cr(acac)2(ox)]-

```
(acac- = acetylacetonate, ox2- = oxalate) in aq. soln. afforded the novel
     3d-4f heterodinuclear complexes [(acac)2Cr(ox)Ln(HBpz3)2] (Ln = Eu (1), Gd
     (2), Tb (3), Yb (4), Lu (5)). 4 Crystallizes in monoclinic space group
    P2/n, with a 8.594(3), b 18.538(4), c 12.093(2) .ANG., .beta.
    93.71(2).degree., and Z = 2. Yb coordinates in an eight-coordinate
    distorted square antiprismatic geometry. The intramol.
    Cr.cntdot..cntdot..yb distance is 5.631(1) .ANG.. The magnetic
    susceptibility data for 2 showed that the CrIII-GdIII interaction is
    weakly antiferromagnetic with an exchange coupling const. JCrGd = -0.09
           The luminescence measurements demonstrated the energy
    transfers for both Ln(III) .fwdarw. Cr(III) and Cr(III) .fwdarw. Ln(III),
    of which the degree of emission quenching depends on the energy gap of the
    excited levels in two metal centers. These results reveal that the
    metal-metal interactions between Cr(III) and Ln(III) are very weak in
    magnetic interaction but are strong from the viewpoint of energy transfer.
ST
    crystal structure lanthanide chromium oxalato pyrazolylborate; rare earth
    pyrazolylborate chromium oxalato prepn; pyrazolylborate lanthanide oxalato
    bridged chromium prepn; antiferromagnetic exchange gadolinium
    pyrazolylborate chromium oxalato; magnetic susceptibility lanthanide
    pyrazolylborate chromium oxalato; phosphorescence lanthanide
    pyrazolylborate chromium acac oxalato; quenching phosphorescence europium
    terbium chromium oxalato
    Phosphorescence quenching
IT
        (in europium and terbium hydrotris(pyrazol-1-yl)borate chromium acac
       oxalato bridged heterodinuclear complex)
TT
    Antiferromagnetic exchange
        (of gadolinium hydrotris(pyrazol-1-yl)borate chromium acac oxalato
       bridged heterodinuclear complex)
    Magnetic susceptibility
TΥ
    Phosphorescence
        (of rare earth hydrotris(pyrazol-1-yl)borate chromium acac oxalato
       bridged heterodinuclear complex)
ΙT
    Crystal structure
    Molecular structure
        (of ytterbium hydrotris(pyrazol-1-yl)borate chromium acac oxalato
       bridged heterodinuclear complex)
TΤ
    Rare earth carboxylic acid complexes
    RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn., magnetic susceptibility and phosphorescence)
IT
    18583-60-3, Potassium hydrotris(pyrazol-1-yl)borate
    RL: RCT (Reactant)
        (complexation with rare earth ions)
IT
    583-52-8, Dipotassium oxalate
    RL: RCT (Reactant)
        (for prepn. of chromium acac oxalate complex)
ΙT
     212889-82-2P
                    212889-85-5P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and magnetic susceptibility)
IT
     212889-84-4P
    RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and magnetic susceptibility and crystal structure)
IT
     212889-86-6P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and reaction with hydrotris(pyrazol-1-yl)borate and rare earth
        ions)
IT
     212889-81-1P 212889-83-3P
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN
     (Synthetic preparation); PREP (Preparation); PROC (Process)
        (prepn., magnetic susceptibility and phosphorescence quenching)
     212889-82-2P
TΤ
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RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and magnetic susceptibility)

RN 212889-82-2 HCAPLUS

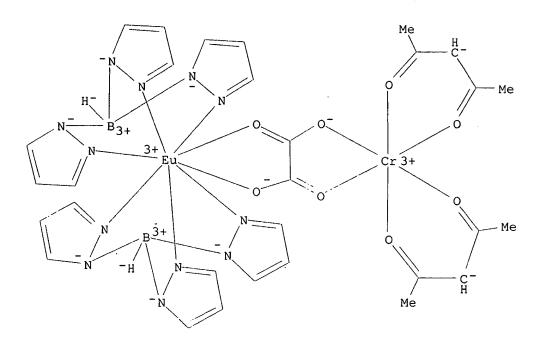
CN Gadolinium, [bis(2,4-pentanedionato-.kappa.O,.kappa.O')chromium][.mu.[ethanedioato(2-)-.kappa.O1,.kappa.O2':.kappa.O1',.kappa.O2]]bis[hydrotris(1H-pyrazolato-.kappa.N1)borato(1-)-.kappa.N2,.kappa.N2',.kappa.N2''](9CI) (CA INDEX NAME)

IT 212889-81-1P 212889-83-3P

RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (prepn., magnetic susceptibility and phosphorescence quenching)

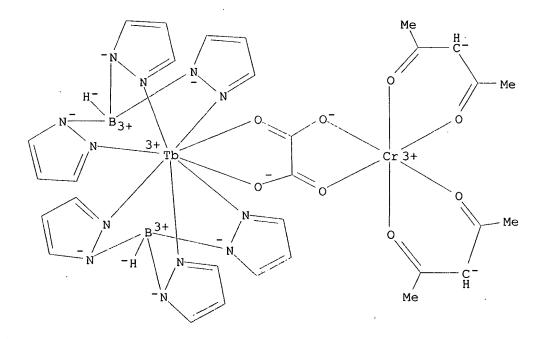
RN 212889-81-1 HCAPLUS

CN Europium, [bis(2,4-pentanedionato-.kappa.O,.kappa.O')chromium][.mu.[ethanedioato(2-)-.kappa.O1,.kappa.O2':.kappa.O1',.kappa.O2]]bis[hydrotris
(1H-pyrazolato-.kappa.N1)borato(1-)-.kappa.N2,.kappa.N2',.kappa.N2''](9CI) (CA INDEX NAME)



RN 212889-83-3 HCAPLUS

CN Terbium, [bis(2,4-pentanedionato-.kappa.O,.kappa.O')chromium][.mu.[ethanedioato(2-)-.kappa.O1,.kappa.O2':.kappa.O1',.kappa.O2]]bis[hydrotris
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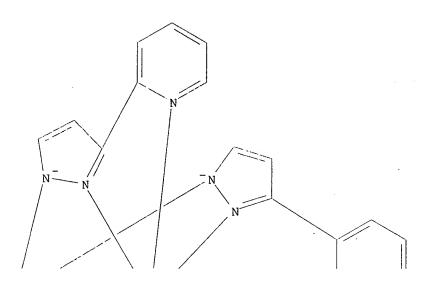


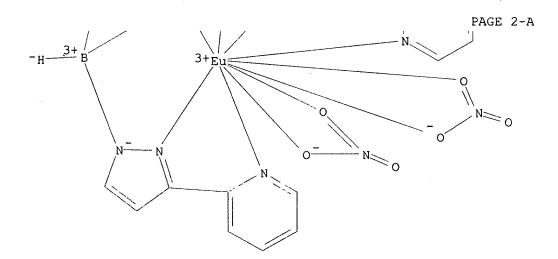
L16 ANSWER 10 OF 17 HCAPLUS COPYRIGHT 2002 ACS

AN 1997:646437 HCAPLUS

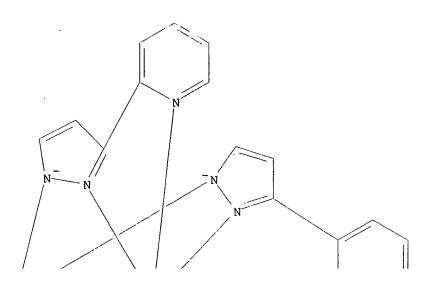
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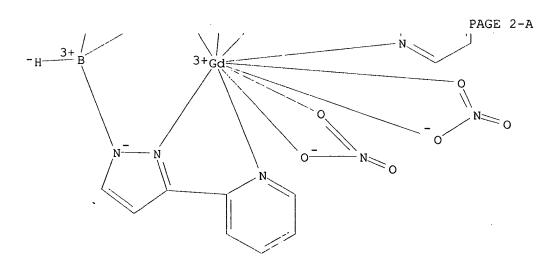
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Luminescence properties of Eu3+, Tb3+, and Gd3+ complexes of the
TΙ
     hexadentate N-donor podand tris-[3-(2-pyridyl)pyrazol-1-yl]hydroborate
     Armaroli, Nicola; Balzani, Vincenzo; Barigelletti, Francesco; Ward,
ΑU
     Michael D.; McCleverty, Jon A.
CS
     Istituto di Fotochimica e Radiazioni d'Alta Energia del CNR, Via Gobetti
     101, Bologna, 40129, Italy
Chem. Phys. Lett. (1997), 276(5,6), 435-440
SO
     CODEN: CHPLBC; ISSN: 0009-2614
PB
     Elsevier
DT
     Journal
LA
     English
     73-5 (Optical, Electron, and Mass Spectroscopy and Other Related
CC
     Properties)
AB
     The photophys. properties of Gd3+, Eu3+, and Tb3+ complexes of a podand
     ligand tris[3-(2-pyridyl)-pyrazolyl]hydroborate (L-) were studied in 5
     solvents. [Gd(L)(NO3)2] (Gd-A) exhibits only the typical ligand-centered
     spectroscopy, i.e. luminescence and triplet-triplet absorption,
     whereas Eu-A and Tb-A show only the characteristic metal centered
     luminescence. The podand does not fully surround the metal ion
     and the ligand-to-metal energy transfer efficiency is high for Tb-A
     because a large energy gap prevents back energy transfer, but it is small
     for Eu-A, because of the presence of low energy charge-transfer excited
     states. Also the 3 bis-podand complexes [LnL2]+ (Ln-B) were studied, but
     only in CH2Cl2 for soly. reasons.
    luminescence lanthanide hexadentate podand
ST
     pyridylpyrazolylhydroborate
IΤ
    Band gap
    Electron transfer
     Energy transfer
     Excited state
       Luminescence
     Triplet state
        (of lanthanide complexes of hexadentate N-donor podand
        tris[(pyridyl)pyrazolyl]hydroborate)
ΙT
     UV and visible spectra
        (transient; of lanthanide complexes of hexadentate N-donor podand
        tris[(pyridyl)pyrazolyl]hydroborate)
     UV and visible spectra
IT
        (triplet-triplet; of lanthanide complexes of hexadentate N-donor podand
        tris[(pyridyl)pyrazolyl]hydroborate)
     Rare earth compounds
IT
     RL: PEP (Physical, engineering or chemical process); PRP (Properties);
     PROC (Process)
        (tris[(pyridyl)pyrazolyl]hydroborate complexes; luminescence
IT
     185199-01-3 185199-02-4 185199-03-5
     185199-08-0 185199-10-4 185199-12-6
     RL: PEP (Physical, engineering or chemical process); PRP (Properties);
     PROC (Process)
        (luminescence of)
ΙT
     185199-01-3 185199-02-4 185199-03-5
     185199-08-0 185199-10-4 185199-12-6
     RL: PEP (Physical, engineering or chemical process); PRP (Properties);
     PROC (Process)
        (luminescence of)
     185199-01-3 HCAPLUS
RN
     Europium, [hydrotris[2-(1H-pyrazol-3-yl-.kappa.Nl)pyridinato]borato(1-
CN
     )]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)
```





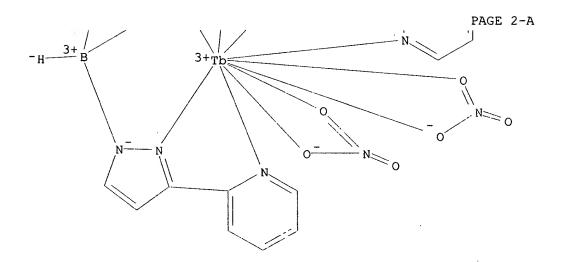
RN 185199-02-4 HCAPLUS CN Gadolinium, [hydrotris[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)





RN 185199-03-5 HCAPLUS

CN Terbium, [hydrotris[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)



RN

185199-08-0 HCAPLUS Europium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]-, tetraphenylborate(1-) (9CI) (CA INDEX NAME) CN

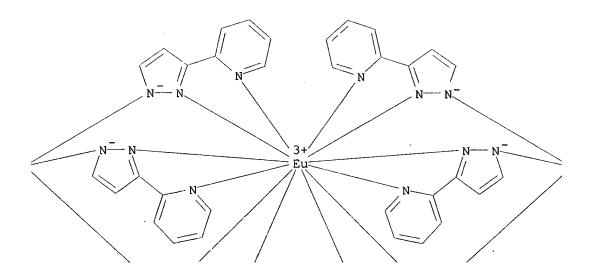
CM 1

CRN 185199-07-9

CMF C48 H38 B2 Eu N18

CCI CCS

PAGE 1-B



PAGE 1-C

}B 3+ H-

PAGE 2-B

CM 2

CRN 4358-26-3 CMF C24 H20 B CCI CCS

RN 185199-10-4 HCAPLUS

Gadolinium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]-,
tetraphenylborate(1-) (9CI) (CA INDEX NAME)

CM 1

0.9/445050 ΧU Page 102

CRN CMF CCI

185199-09-1 C48 H38 B2 Gd N18 CCS

PAGE 1-A

PAGE 1-B

PAGE 1-C

B 3+ H-

PAGE 2-B

CM 2

CRN 4358-26-3 CMF C24 H20 B CCI CCS

$$\begin{array}{c|c}
\hline
 & C \\
\hline
 & C \\
\hline
 & B \\
\hline
 & C \\
 & C \\
\hline
 & C \\
 & C$$

RN

185199-12-6 HCAPLUS
Terbium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]-, tetraphenylborate(1-) (9CI) (CA INDEX NAME)

CM 1

CN

XU 09/445050 Page 104

CRN 185199-11-5 CMF C48 H38 B2 N18 Tb CCI CCS

PAGE 1-A

PAGE 1-B

PAGE 1-C

B 3+ H-

CM

CRN 4358-26-3 CMF C24 H20 B CCI CCS

$$\begin{array}{c|c}
\hline
 & & \\
\hline$$

L16 ANSWER 11 OF 17 HCAPLUS COPYRIGHT 2002 ACS AN 1997:594503 HCAPLUS

127:240696 DN

TΙ Fluorescent compounds

Bell, Colin David; Howse, John Hewer Coles; Bosworth, Nigel; James, David IN

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

Martin

Amersham International PLC, UK PA

U.S., 30 pp. Cont.-in-part of U. S. 5,435,937. SO

CODEN: USXXAM

DT Patent English LA

ICM C09K011-04 IC

ICS C09K011-06; C07F009-535

NCL 252301180

CC 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 29, 71

FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 5658494	Α	19970819	US 1995-445858	19950522
	US 5435937	Α	19950725	US 1993-17674	19930212
	CA 2176525	AA	19961123	CA 1996-2176525	19960514
PRAI	EP 1992-301249		19920214		
	US 1993-17674		19930212		
	US 1995-445858		19950522		

MARPAT 127:240696 OS

Radioluminescent bodies are described which comprise a polymer AΒ together with a chelate of a transition or lanthanide or actinide metal ion, which body is transparent or translucent, wherein the body is radioactively labeled with tritium and has the property of emitting light or IR radiation by virtue of internally generated ionizing radiation resulting from radioactive decay of the tritium. Fluorescent body composed of a polymer together with a chelate of a transition or lanthanide or actinide metal ion, which body is transparent or translucent and has the property of emitting light or IR radiation when subjected to UV or ionizing radiation are also described wherein there is present a siloxane which improves the stability and light output or a free radical scavenger which reduces polymer degrdn. The compd. that results from reacting p-tolyldiphenylphosphine oxide with trivalent terbium tris(dipivaloyl methide) (sic) is also claimed.

STchelate fluorescent radioluminescent compd

IΤ Fluorescent substances Scintillators

(fluorescent and radioluminescent compds. and compns.)

ΙT Luminescent substances

> (radio-; fluorescent and radioluminescent compds. and compns.)

7440-53-1DP, Europium, ΙT 893-33-4DP, europium bipyridine complex bipyridine trifluoro(naphthyl)butanedione complex 20219-51-6P 37275-48-2DP, Bipyridine, europium trifluoro(naphthyl)butanedione complex 156882-92-7P 156915-46-7P 156915-57-0P 156952-11-3P 156952-12-4P 188293-81-4P 156952-13-5P 185448-10-6P 195316-76-8P Tetrakis(2,2,6,6-tetramethyl-3,5-heptanedionato)terbium(III) piperidine

RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(fluorescent and radioluminescent compds. and compns.)

33134-19-9 33134-20-2 36836-37-0 55188-26-6 131006-98-9 TΤ 156882-91-6 156915-48-9 156915-49-0 156915-50-3 156915-51-4 156915-52-5 156915-53-6 156915-54-7 156915-55-8 156915-56-9 156952-10-2 156952-09-9

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

```
(fluorescent and radioluminescent compds. and compns.)
     110-89-4, Piperidine, reactions 120-46-7, Dibenzoyl methane
IT
                                                                     541-09-3,
                     603-35-0, Triphenylphosphine, reactions
     Uranyl acetate
                                                              893-33-4
     999-97-3, Hexamethyldisilazane 1031-93-2, (4-Methylphenyl)diphenyl
     phosphine
                1073-67-2, P-Chlorostyrene 1118-71-4, 2,2,6,6-Tetramethyl-
     3,5-heptanedione 1499-21-4, Diphenyl-phosphinic acid chloride
     6840-28-4, p-Tolyldiphenylphosphine oxide
                                               10025-76-0, Europium
                   10361-82-7, Samarium chloride
                                                   15492-51-0,
     trichloride
     Tris(2,2,6,6-tetramethyl-3,5-heptanedionato)terbium
                                                          15522-69-7,
     Tris(2,2,6,6-tetramethyl-3,5-heptanedionato)dysprosium(III)
                                                                  26628-22-8,
                    37275-48-2, Bipyridyl
     Sodium azide
     RL: RCT (Reactant)
        (fluorescent and radioluminescent compds. and compns.)
     2960-37-4P
                  4129-17-3P, Diphenylphosphinic azide
ΙT
                                                       14552-07-9P
     15492-50-9P
                  19269-14-8P
                                 24082-36-8P, Diphenyl-phosphonimido-triphenyl
                   31239-06-2P
    phosphorane
                                 40538-11-2P, p-Styryldiphenylphosphine
     47182-95-6P
                  80233-27-8P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
        (fluorescent and radioluminescent compds. and compns.)
ΙT
     188293-72-3P
     RL: SPN (Synthetic preparation); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (fluorescent and radioluminescent compds. and compns.)
     119-64-2, Tetralin.
                          3390-61-2, Pentaphenyltrimethyltrisiloxane.
TΤ
     3982-82-9
                 9003-53-6, Polystyrene
                                         9003-53-6D, Polystyrene, tritiated
     9017-21-4, Polyvinyltoluene
                                 10028-17-8, Tritium, uses
                                                              18666-24-5D,
     Ethyltriphenylsilane, tritiated
                                      24936-41-2, 4-Methylstyrene homopolymer
     24936-44-5, 4-Methoxystyrene homopolymer 25232-08-0, 4-Vinyl-biphenyl
                   25990-16-3, 2,4-Dimethylstyrene homopolymer
                                                                26009-55-2,
    homopolymer
                                    27756-35-0, 2,4,6-Trimethylstyrene
     4-t-Butylstyrene homopolymer
    homopolymer
                  62125-00-2D, Diphenylethylbenzene., tritiated
     RL: TEM (Technical or engineered material use); USES (Uses)
        (fluorescent and radioluminescent compds. and compns.)
IT
    20219-51-6P
     RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (fluorescent and radioluminescent compds. and compns.)
     20219-51-6 HCAPLUS
RN
     Terbium, tris(2,2,6,6-tetramethyl-3,5-heptanedionato-
CN
     .kappa.O,.kappa.O')[1,1',1''-borylidynetris[1H-pyrazole-.kappa.N2]]- (9CI)
       (CA INDEX NAME)
```

L16 ANSWER 12 OF 17 HCAPLUS COPYRIGHT 2002 ACS

AN 1997:441000 HCAPLUS

DN 127:144291

TI Lanthanide complexes of the tetradentate N-donor ligand dihydrobis[3-(2-pyridyl)pyrazolyl]borate and the terdentate N-donor ligand 2,6-bis(1H-pyrazol-3-yl)pyridine: syntheses, crystal structures and solution structures based on luminescence lifetime studies

AU Bardwell, David A.; Jeffery, John C.; Jones, Peter L.; McCleverty, Jon A.; Psillakis, Elefteria; Reeves, Zoe; Ward, Michael D.

CS School of Chemistry, University of Bristol, Bristol, BS8 1TS, UK

SO J. Chem. Soc., Dalton Trans. (1997), (12), 2079-2086 CODEN: JCDTBI; ISSN: 0300-9246

PB Royal Society of Chemistry

DT Journal

LA English

CC 78-7 (Inorganic Chemicals and Reactions) Section cross-reference(s): 75

Lanthanide complexes of two polydentate N-donor ligands contg. a mixt. of AB pyridyl and pyrazolyl donors were prepd. Dihydrobis[3-(2pyridyl)pyrazolyl]borate (L1) - is a tetradentate ligand with two bidentate chelating pyridyl/pyrazolyl arms linked by an apical BH2 group; 2,6-bis(1H-pyrazol-3-yl)pyridine (L2) is a terdentate chelating ligand reminiscent of terpyridine. Reaction of L1 with lanthanide salts gave [M(L1)2X]n+; the crystal structures of [Eu(L1)2(DMF)][ClO4].cntdot.2.5CH2C 12, [Tb(L1)2(NO3)].cntdot.2CH2Cl2 and [Tb(L1)2(H2O)][L1].cntdot.H2O.cntdot .0.5CH2Cl2 were detd. and all contain two tetradentate ligands L1 and an ancillary ligand X [DMF, nitrate or water] whose nature depends on the reaction/recrystn. conditions to complete the coordination sphere. Luminescence studies of [Tb(L1)2(NO3)] in water or D2O and MeOH or CD3OD showed that in methanol the solvation no. q is .apprxeq. 1.8, consistent with displacement of nitrate by the solvent; however in water q .apprxeq. 4.5, indicating addnl. displacement of some of the N-donor heterocyclic rings of L1 by coordinating water mols. Reaction of L2 with lanthanide salts afforded [M(L2)3]3+, all isolated as their hexafluorophosphate salts. The crystal structures of three of these (M = 1) Eu, Gd or Ho) showed that they are isostructural and isomorphous, with tricapped trigonal-prismatic nine-coordinate geometries similar to that of [M(terpy)3]3+ (terpy = 2,2':6',2''-terpyridine). Luminescence studies of [Tb(L2)3][PF6]3 gave a solvation no. q of 0.6 in methanol,

which is small enough to be accounted for by 2nd-sphere solvation effects alone and therefore suggests that the nine-coordinate structure is retained in methanol soln. However, in water, q is again .apprxeq. 4.5, due to displacement of some of the donor groups of the L2 ligands by water.

ST crystal structure rare earth dihydrobispyridylpyrazolylborato bispyrazolylpyridine; rare earth dihydrobispyridylpyrazolylborato bispyrazolylpyridine prepn structure; hydrobispyridylpyrazolylborate europium terbium complex prepn structure; pyridylpyrazolylborate europium terbium complex prepn structure; pyrazolylpyridine rare earth complex prepn structure; terbium dihydrobispyridylpyrazolylborato bispyrazolylpyridine prepn structure solvation; luminescence lifetime terbium dihydrobispyridylpyrazolylborato bispyrazolylpyridine; solvation number terbium dihydrobispyridylpyrazolylborato bispyrazolylpyridine

IT Excited state

(lifetime; soln. structure of terbium dihydrobis(pyridylpyrazolyl)borat o and bis(pyrazolyl)pyridine complexes based on luminescence lifetime)

IT Crystal structure

Molecular structure

(of rare earth dihydrobis(pyridylpyrazolyl)borato and bis(pyrazolyl)pyridine complexes)

IT Rare earth complexes

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and crystal and soln. structures of rare earth dihydrobis(pyridylpyrazolyl)borato and bis(pyrazolyl)pyridine complexes)

IT Solvation number

(soln. structure of terbium dihydrobis(pyridylpyrazolyl)borato and bis(pyrazolyl)pyridine complexes based on luminescence lifetime)

IT 63285-53-0, 2,6-Bis(1H-pyrazol-3-yl)pyridine

RL: RCT (Reactant)

(for prepn. of rare earth bis(pyrazolyl)pyridine complexes)

IT 13762-51-1, Potassium tetrahydroborate 75415-03-1, 3-(2-Pyridyl)pyrazole RL: RCT (Reactant)

(for prepn. of rare earth dihydrobis(pyridylpyrazolyl)borato complexes)

IT 192885-45-3P, Potassium dihydrobis[3-(2-pyridyl)pyrazolyl]borate
 RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)

(for prepn. of rare earth dihydrobis(pyridylpyrazolyl)borato complexes)

IT 192885-28-2P 192885-30-6P 192885-35-1P

192885-37-3P 192885-39-5P 192885-43-1P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and crystal structure)

IT 192885-27-1P 192885-34-0P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and mol. structure)

IT 192885-41-9P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., luminescence lifetimes and soln. structures)

IT 192885-29-3P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., mol. and soln. structures and luminescence lifetimes)

IT 192885-28-2P 192885-30-6P 192885-35-1P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and crystal structure)

RN 192885-28-2 HCAPLUS

CN Europium(1+), bis[dihydrobis[2-(1H-pyrazol-3-yl-

.kappa.N1)pyridinato]borato(1-)](N,N-dimethylformamide-.kappa.O)-, perchlorate, compd. with dichloromethane (2:5) (9CI) (CA INDEX NAME)

1 CM

CRN 75-09-2 CMF C H2 C12

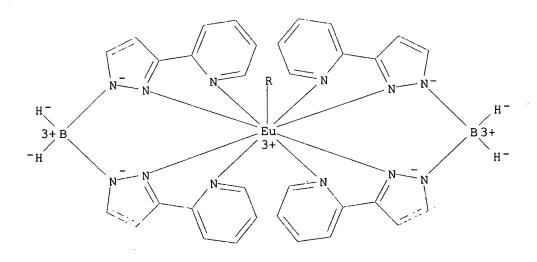
 ${\tt Cl-CH_2-Cl}$

CM 2

CRN 192885-27-1 CMF C35 H35 B2 Eu N13 O . Cl O4

> CM 3

CRN 192885-26-0 CMF C35 H35 B2 Eu N13 O CCI CCS



 $Me_2N-CH=0$

CM

CRN 14797-73-0 CMF Cl O4

RN 192885-30-6 HCAPLUS

CN Terbium, bis[dihydrobis[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-)](nitrato-.kappa.O,.kappa.O')-, compd. with dichloromethane (1:2) (9CI) (CA INDEX NAME)

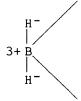
CM 1

CRN 192885-29-3

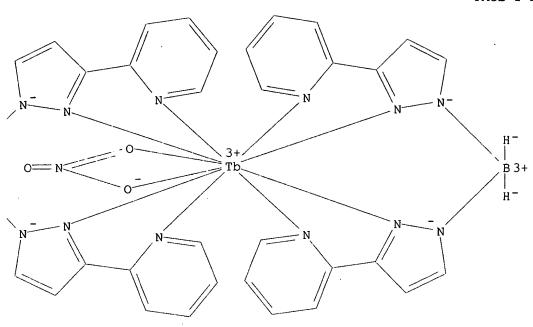
CMF C32 H28 B2 N13 O3 Tb

CCI CCS

PAGE 1-A



PAGE 1-B



CM 2

CRN 75-09-2 CMF C H2 C12

 ${\tt Cl-CH_2-Cl}$

RN 192885-35-1 HCAPLUS

CN Terbium(1+), aquabis[dihydrobis[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-)]-, (T-4)-dihydrobis[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borate(1-), compd. with dichloromethane (2:1), dihydrate (9CI) (CA INDEX NAME)

CM 1

CRN 75-09-2 CMF C H2 C12

C1-CH2-C1

CM 2

CRN 192885-33-9

CMF $\,$ C32 $\,$ H30 $\,$ B2 $\,$ N12 $\,$ O $\,$ Tb $\,$. C16 $\,$ H14 $\,$ B $\,$ N6

CM 3

XU 09/445050 Page 113

CRN 192885-32-8 CMF C16 H14 B N6 CCI CCS CDES 7:T-4

$$\begin{array}{c|c}
N & H^{-} \\
N & B \\
N & H^{-}
\end{array}$$

CM 4

CRN 192885-31-7 CMF C32 H30 B2 N12 O Tb CCI CCS

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

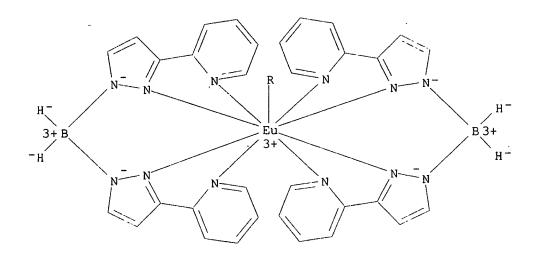
IT 192885-27-1P 192885-34-0P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and mol. structure)

RN 192885-27-1 HCAPLUS

CM 1

CRN 192885-26-0 CMF C35 H35 B2 Eu N13 O CCI CCS



$$\begin{array}{c} \text{Me2N-CH-O} \\ | \\ | \\ R \end{array}$$

CRN 14797-73-0 CMF Cl O4

RN 192885-34-0 HCAPLUS

CN Terbium(1+), aquabis[dihydrobis[2-(1H-pyrazol-3-yl-

- .kappa.N1)pyridinato]borato(1-)]-, (T-4)-dihydrobis[2-(1H-pyrazol-3-yl-
- .kappa.N1)pyridinato]borate(1-), monohydrate (9CI) (CA INDEX NAME)

CM 1

192885-33-9 CRN

CMF C32 H30 B2 N12 O Tb . C16 H14 B N6

> 2 . CM

192885-32-8 CRN CMF C16 H14 B N6 CCI CCS

CDES 7:T-4

$$\begin{array}{c|c} N & H^{-} \\ N & H^{-} \end{array}$$

CRN 192885-31-7

CMF C32 H30 B2 N12 O Tb

CCI CCS

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

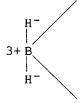
IT 192885-29-3P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., mol. and soln. structures and luminescence lifetimes)

RN 192885-29-3 HCAPLUS

CN Terbium, bis[dihydrobis[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-)](nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)

PAGE 1-A



PAGE 1-B

ANSWER 13 OF 17 HCAPLUS COPYRIGHT 2002 ACS

1996:734439 HCAPLUS AN

DN 126:69292

TI Lanthanide Complexes of the Hexadentate N-donor Podand Tris[3-(2-pyridyl)pyrazolyl]hydroborate: Solid-State and Solution Properties

ΑU Jones, Peter L.; Amoroso, Angelo J.; Jeffery, John C.; McCleverty, Jon A.; Psillakis, Elefteria; Rees, Leigh H.; Ward, Michael D. School of Chemistry, University of Bristol, Bristol, BS8 1TS, UK Inorg. Chem. (1997), 36(1), 10-18 CODEN: INOCAJ; ISSN: 0020-1669

CS

SO

PB American Chemical Society

DTJournal

LA English

78-7 (Inorganic Chemicals and Reactions) CC Section cross-reference(s): 8, 75

The hexadentate N6-donor podand tris[3-(2-pyridyl)pyrazolyl]hydroborate AB (TpPy) contains 2-pyridyl fragments attached to the pyrazolyl C3-positions such that each arm is a bidentate chelate. Three series of lanthanide(III) complexes were prepd.: [M(TpPy)(MeOH)2F][PF6] (series A), [M(TpPy)(NO3)2] (series B), and [M(TpPy)2][BPh4] (series C). Crystallog. studies showed that series A and B have a 1:1 metal: TpPy ratio, with the metal ion lying within the podand cavity and the remaining coordination sites occupied by solvent mols. and/or counterions to give 9-coordination (A, with one fluoride and two MeOH ligands) or 10-coordination (B, with two bidentate nitrate ligands). The C complexes were prepd. in the absence of any coordinating anions and have a 1:2 metal: TpPy ratio with an unusual icosahedral geometry arising from coordination of the 12 N donors from two interleaved podands. Soln. cond. studies on the B complexes show that in H2O the nitrates dissoc. to give [M(TpPy)(H2O)q](NO3)2; the relaxivity of [Gd(TpPy)(NO3)2] in H2O is 4.4 s-1 mM-1, a value comparable to those of clin. useful MRI contrast enhancement agents. Comparison of

ST

TT

IT

IT

ΙT

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IT

ΙT

TΤ

IT

ΙT

IT

IT

ΙT

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09/445050
                  Page 117
emission lifetimes of [M(TpPy)(NO3)2] (M = Eu, Tb) in H2O/D2O and
MeOH/CD3OD give values for q, the no. of coordinated solvent mols., of 3.6
(water) and 2.6 (MeOH). The C complex [Tb(TpPy)2][BPh4] also has q = 2.6
in MeOH, suggesting that partial ligand dissocn. allows access of solvent
mols. to the metal coordination sphere.
lanthanide trispyridylpyrazolylhydroborate complex prepn crystal
structure; pyridylpyrazolylhydroborate tris podand ligand lanthanide
prepn; luminescence lanthanide trispyridylpyrazolylhydroborate;
MRI contrast agent prototype lanthanide trispyridylpyrazolylhydroborate;
coordination trispyridylpyrazolylhydroborate podand lanthanide; NMR
relaxivity lanthanide trispyridylpyrazolylhydroborate
MRI (magnetic resonance imaging)
   (lanthanide tris[3-(2-pyridyl)pyrazolyl]hydroborate complexes as
   prototype MRI contrast enhancement agents)
Coordination number
Crystal structure
  Luminescence
Molecular structure
   (of lanthanide tris[3-(2-pyridyl)pyrazolyl]hydroborate complexes)
Rare earth complexes
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
   (prepn. and structure of lanthanide tris[3-(2-
   pyridyl)pyrazolyl]hydroborate complexes)
Podands
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
   (prepn. of tris[3-(2-pyridyl)pyrazolyl]hydroborate podand ligand for
   lanthanides)
161095-31-4P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
   (for prepn. of lanthanide tris[3-(2-pyridyl)pyrazolyl]hydroborate
   complexes)
13762-51-1, Potassium tetrahydroborate 75415-03-1, 3-(2-Pyridyl)pyrazole
RL: RCT (Reactant)
   (for prepn. of tris[3-(2-pyridyl)pyrazolyl]hydroborate ligand)
185199-00-2P 185199-02-4P 185199-04-6P
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
   (prepn. and NMR relaxivity of)
171198-27-9P 185199-13-7P
                            185199-16-0P
185199-18-2P 185199-20-6P 185199-24-0P
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
   (prepn. and crystal structure of)
161095-30-3P 171198-26-8P
                           185198-99-6P
185199-08-0P
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
   (prepn. and mol. structure of)
185198-89-4P 185198-91-8P 185198-93-0P
185198-95-2P
               185198-97-4P
                             185198-98-5P
                                             185199-06-8P
185199-10-4P
RL: SPN (Synthetic preparation); PREP (Preparation)
   (prepn. of)
185199-03-5P
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
   (prepn., luminescence, NMR relaxivity, and soln. cond. of)
185199-12-6P
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
   (prepn., luminescence, and soln. cond. of)
185199-01-3P
```

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., mol. structure, luminescence, NMR relaxivity, and

soln. cond. of)

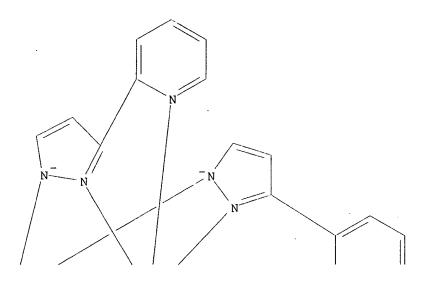
XU 09/445050 Page 118

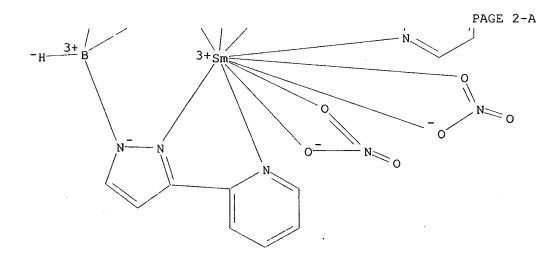
IT 185199-00-2P 185199-02-4P 185199-04-6P

RN 185199-00-2 HCAPLUS

CN Samarium, [hydrotris[2-(lH-pyrazol-3-yl-.kappa.N1)pyridinato]borato(l-)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)

PAGE 1-A

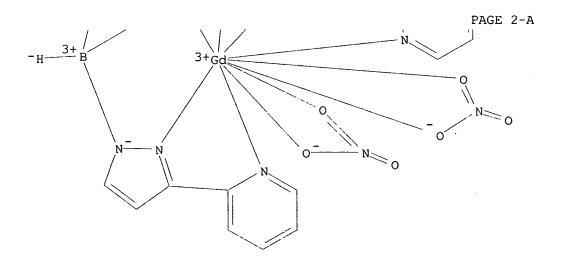




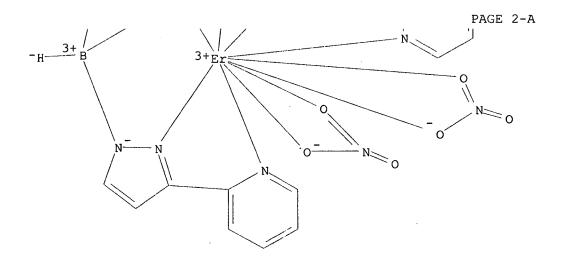
RN 185199-02-4 HCAPLUS

CN Gadolinium, [hydrotris[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)

PAGE 1-A



RN 185199-04-6 HCAPLUS Erbium, [hydrotris[2-(1H-pyrazol-3-yl-.kappa.Nl)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME) CN



RN 171198-27-9 HCAPLUS

CN Samarium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-)]-, tetraphenylborate, compd. with 1,1'-oxybis[ethane] (1:1) (9CI) (CA INDEX NAME)

CM 1

XU 09/445050 Page 121

CRN 60-29-7 CMF C4 H10 O

H3C-СH2-О-СH2-СН3

CM 2

CRN 171198-26-8 CMF C48 H38 B2 N18 Sm . C24 H20 B

> CM 3

CRN 171198-25-7

CMF C48 H38 B2 N18 Sm CCI CCS

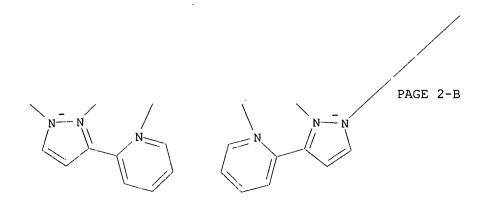
PAGE 1-A

-H-3+B

PAGE 1-B

PAGE 1-C

≥B 3+ H-



CRN 4358-26-3 CMF C24 H20 B CCI CCS

RN 185199-13-7 HCAPLUS

CN Europium, fluoro[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]bis(methanol)-, hexafluorophosphate(1-), compd. with methanol and 1,1'-oxybis[ethane] (3:2:2), tetrahydrate (9CI) (CA INDEX NAME)

CM 1

CRN 67-56-1 CMF C H4 O

нзс-он

CM 2

CRN 60-29-7 CMF C4 H10 O

H3C-СH2-О-СH2-СН3

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

CRN 161095-30-3

CMF C26 H27 B Eu F N9 O2 . F6 P

CM 4

CRN 161095-29-0

CMF C26 H27 B Eu F N9 O2 CCI CCS

PAGE 1-A

PAGE 2-A



CM 5

CRN 16919-18-9 CMF F6 P

CCI CCS

RN 185199-18-2 HCAPLUS

CN Europium, [hydrotris{2-(lH-pyrazol-3-yl)pyridinato]borato(l-)]bis(nitrato-.kappa.O,.kappa.O')-, compd. with N,N-dimethylformamide and 1,1'-oxybis[ethane] (2:2:1) (9CI) (CA INDEX NAME)

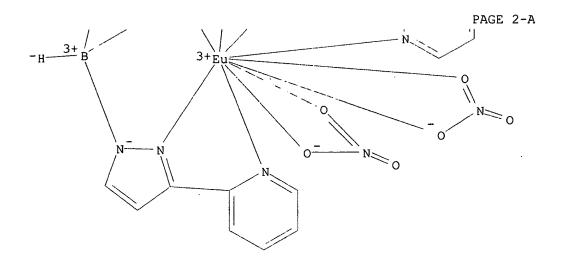
CM 1

CRN 185199-01-3

CMF C24 H19 B Eu N11 O6

CCI CCS

PAGE 1-A



CRN 68-12-2 CMF C3 H7 N O

H3C-N-CH=0

CM 3

CRN 60-29-7 CMF C4 H10 O

H3C-CH2-O-CH2-CH3

RN 185199-20-6 HCAPLUS

CN Erbium, [hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]bis(nitrato-.kappa.O,.kappa.O')-, compd. with N,N-dimethylformamide and 1,1'-oxybis[ethane] (2:2:1) (9CI) (CA INDEX NAME)

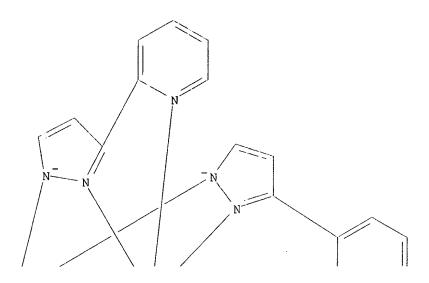
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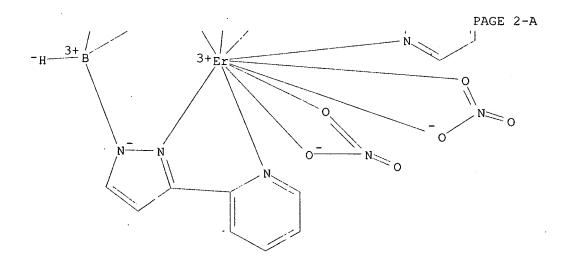
CRN 185199-04-6

CMF C24 H19 B Er N11 O6

CCI CCS

PAGE 1-A





CM 2

CRN 68-12-2 CMF C3 H7 N O

```
CH3
H3C-N-CH-0
```

CRN 60-29-7 CMF C4 H10 O

H3C-CH2-O-CH2-CH3

185199-24-0 HCAPLUS RN

Europium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]-, CN tetraphenylborate(1-), compd. with 1,1'-oxybis[ethane] (1:1) (9CI) (CA INDEX NAME)

CM1

CRN 60-29-7 CMF C4 H10 O

H3C-CH2-O-CH2-CH3

2 CM

CRN 185199-08-0

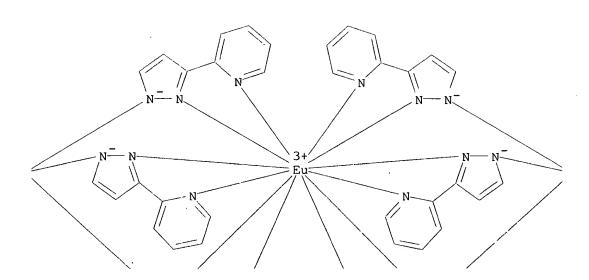
CMF C48 H38 B2 Eu N18 . C24 H20 B

CM 3

CRN 185199-07-9

CMF C48 H38 B2 Eu N18 CCI CCS

PAGE 1-B



PAGE 1-C

§ B 3+ H-

PAGE 2-B

CM 4

CRN 4358-26-3 CMF C24 H20 B CCI CCS

RN

IT 161095-30-3P 171198-26-8P 185199-08-0P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and mol. structure of)

161095-30-3 HCAPLUS

CN Europium(1+), fluoro[hydrotris[2-(1H-pyrazol-3-yl-

.kappa.N1)pyridinato]borato(1-)]bis(methanol)-, hexafluorophosphate(1-)

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(9CI) (CA. INDEX NAME)

CM 1

CRN 161095-29-0

CMF C26 H27 B Eu F N9 O2

CCI CCS

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PAGE 2-A



CM 2

CRN 16919-18-9

CMF F6 P CCI CCS

171198-26-8 HCAPLUS RN

CN Samarium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl-:kappa.N1)pyridinato]borato(1-)]-, tetraphenylborate(1-) (9CI) (CA INDEX NAME)

CM 1

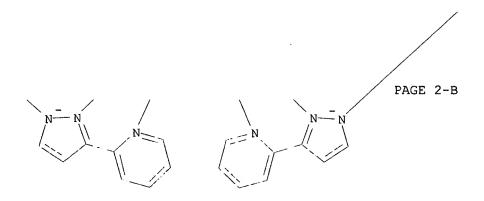
CRN 171198-25-7

CMF C48 H38 B2 N18 Sm

CCI CCS

PAGE 1-A

PAGE 1-C



CRN 4358-26-3 CMF C24 H20 B CCI CCS

RN

185199-08-0 HCAPLUS Europium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]-, tetraphenylborate(1-) (9CI) (CA INDEX NAME) CN

CM 1

CRN 185199-07-9

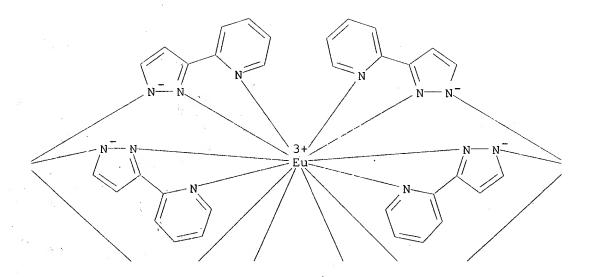
CMF C48 H38 B2 Eu N18

CCI CCS

ΧU

-H-3+

PAGE 1-B



ΧU

PAGE 1-C

B 3+ H-

PAGE 2-B

CM

CRN 4358-26-3 CMF C24 H20 B CCI CCS

$$\begin{array}{c|c}
\hline
C - B & \hline
C - B & \hline
C - C & \hline
\end{array}$$

185198-89-4P 185198-91-8P 185198-93-0P IT

185199-10-4P

RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of)

RN 185198-89-4 HCAPLUS

CN Samarium(1+), fluoro[hydrotris[2-(1H-pyrazol-3-yl-

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

ΧU 09/445050 Page 137

.kappa.N1)pyridinato]borato(1-)]bis(methanol)-, hexafluorophosphate(1-)
(9CI) (CA INDEX NAME)

CM 1

CRN 185198-88-3

CMF C26 H27 B F N9 O2 Sm

CCI CCS

PAGE 1-A

PAGE 2-A

CM 2

CRN 16919-18-9

CMF F6 P

CCI CCS

185198-91-8 HCAPLUS RN

Gadolinium(1+), fluoro[hydrotris[2-(1H-pyrazol-3-yl-CN .kappa.N1)pyridinato]borato(1-)]bis(methanol)-, hexafluorophosphate(1-) (9CI) (CA INDEX NAME)

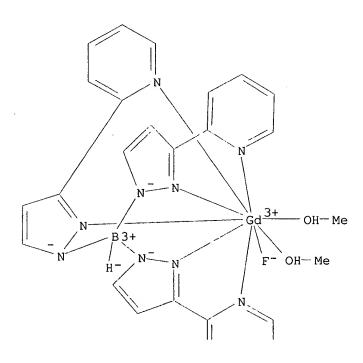
CM 1

CRN 185198-90-7

CMF C26 H27 B F Gd N9 O2

CCI CCS





PAGE 2-A



CRN 16919-18-9

CMF F6 P

RN 185198-93-0 HCAPLUS

CN Terbium(1+), fluoro[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]bis(methanol)-, hexafluorophosphate(1-) (9CI) (CA INDEX NAME)

CM 1

CRN 185198-92-9

CMF C26 H27 B F N9 O2 Tb

CCI CCS

PAGE 1-A

PAGE 2-A

CM 2

CRN 16919-18-9

CMF F6 P

CCI CCS

RN 185199-10-4 HCAPLUS

Gadolinium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]-, tetraphenylborate(1-) (9CI) (CA INDEX NAME) CN

CM 1

CRN 185199-09-1

CMF C48 H38 B2 Gd N18 CCI CCS

PAGE 1-A

PAGE 1-C

PAGE 2-B

CM 2

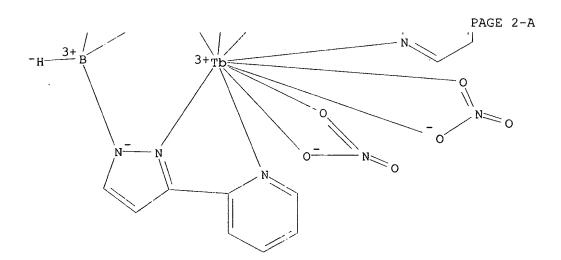
CRN 4358-26-3 CMF C24 H20 B CCI CCS

ΙT 185199-03-5P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., luminescence, NMR relaxivity, and soln. cond. of) 185199-03-5 HCAPLUS

RN

Terbium, [hydrotris[2-(1H-pyrazol-3-yl-.kappa.N1)pyridinato]borato(1-CN)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)



IT 185199-12-6P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., luminescence, and soln. cond. of) 185199-12-6 HCAPLUS

RN

Terbium(1+), bis[hydrotris[2-(1H-pyrazol-3-yl)pyridinato]borato(1-)]-, tetraphenylborate(1-) (9CI) (CA INDEX NAME) CN

CM 1

CRN 185199-11-5

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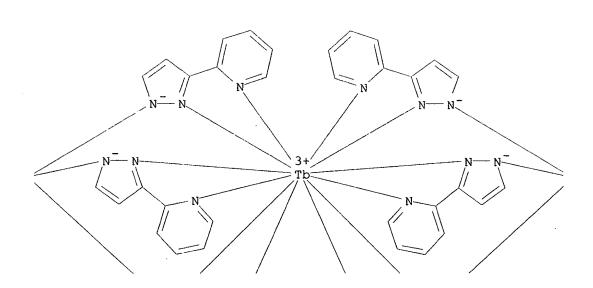
XU 09/445050 Page 144

CMF C48 H38 B2 N18 Tb CCI CCS

PAGE 1-A

-H - 3 + B

PAGE 1-B



} B 3+ H−

PAGE 2-B

CM 2

CRN 4358-26-3 CMF C24 H20 B CCI CCS

IT 185199-01-3P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn., mol. structure, luminescence, NMR relaxivity, and soln. cond. of)

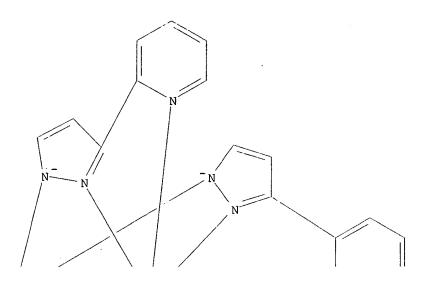
RN 185199-01-3 HCAPLUS

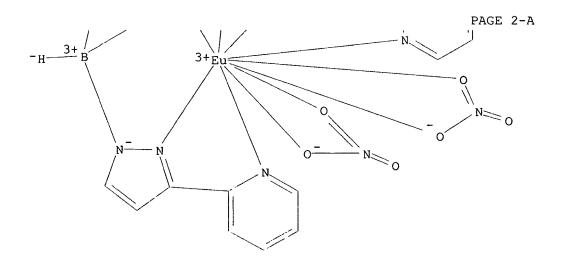
CN Europium, [hydrotris[2-(1H-pyrazol-3-yl-.kappa.Nl)pyridinato]borato(1-

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)]bis(nitrato-.kappa.O,.kappa.O')- (9CI) (CA INDEX NAME)

PAGE 1-A





L16 ANSWER 14 OF 17 HCAPLUS COPYRIGHT 2002 ACS AN 1996:125920 HCAPLUS

124:248762 DN

ΤI $\label{thm:constraints} \mbox{Heteroleptic poly(pyrazolyl)borate derivatives of the lanthanides.}$ Structural and electronic spectral studies of some salicylaldehyde complexes

Lawrence, Royston G.; Jones, Christopher J.; Kresinski, Roman A. ΑU

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Sch. Chem., Univ. Birmingham, Birmingham, B15 2TT, UK CS J. Chem. Soc., Dalton Trans. (1996), (4), 501-7 SO CODEN: JCDTBI; ISSN: 0300-9246 DT Journal LA English 78-7 (Inorganic Chemicals and Reactions) CC Section cross-reference(s): 75 [Ln{HB(pz)3}2L] [pz = pyrazol-1-yl; L = salicylaldehydate, Ln = Y, Pr, Nd, AΒ Sm, Eu, Gd, Tb, Dy, Ho, Er, Yb or Lu; L = 5-methoxysalicylaldehydate (mosal), Ln = Y, Pr, Nd, Sm, Eu, Gd, Tb, Yb or Lu] were synthesized and the crystal structure of [Eu{HB(pz)3}2(mosal)] detd. The Eu ion is eight-coordinate with Eu-O distances of 2.266(5) and 2.402(5) .ANG.; polytopal anal. indicates that the coordination geometry is best described as dodecahedral. The solid-angle sum of 0.768 is close to the norm for eight-coordination. These structural parameters were compared with those calcd. for the previously reported binuclear complex [{Sm[HB(pz)3]2(O2CPh)}2] and estd. for its monomeric counterpart, which is as yet unknown. The use of such data in predicting when complexes of this type will dimerize was assessed. Electronic spectra of the Nd complexes revealed <1% covalency in the metal-ligand bonding and emission spectral data are reported for the Eu and Tb complexes. crystal structure europium pyrazolylborato salicylaldehydato; rare earth ST pyrazolylborato salicylaldehydato prepn Crystal structure TΤ Molecular structure (of europium pyrazolylborato salicylaldehydato complex) ITLuminescence (of rare earth pyrazolylborato salicylaldehydato complexes) ΙT Rare earth compounds RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (pyrazolylborato salicylaldehydato complexes; prepn. and emission spectra of) IT Ultraviolet and visible spectra (emission, of rare earth pyrazolylborato salicylaldehydato complexes) ΙT 90-02-8, Salicylaldehyde, reactions 672-13-9, 5-Methoxysalicylaldehyde 18583-60-3, Potassium hydrotris(1-pyrazolyl)borate RL: RCT (Reactant) (for prepn. of rare earth pyrazolylborato salicylaldehydato complexes) 175091-43-7P IT RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and crystal structure and emission spectra of) 175091-38-0P 175091-39-1P 175091-40-4P IT 175091-41-5P 175091-42-6P RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and emission spectra of) ΙT 175091-23-3P 175091-24-4P 175091-25-5P 175091-26-6P 175091-30-2P 175091-27-7P 175091-28-8P **175091-29-9P** 175091-33-5P 175091-34-6P 175091-31-3P 175091-32-4P 175091-36-8P 175091-37-9P 175091-35-7P RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of) ΙT 175091-43-7P RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(prepn. and crystal structure and emission spectra of) 175091-43-7 HCAPLUS

RN CN Europium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2, N2', N2''](2-hydroxy-5-methoxybenzaldehydato-01,02)-, (DD-8-13233'3'3'3)- (9CI) (CA INDEX NAME)

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

Page 148 XU 09/445050 175091-38-0P 175091-39-1P 175091-40-4P IT 175091-41-5P 175091-42-6P RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and emission spectra of) 175091-38-0 HCAPLUS RN Neodymium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2-CN hydroxybenzaldehydato-O,O')- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** 175091-39-1 HCAPLUS RN Europium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2-CN hydroxybenzaldehydato-O,O')- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** 175091-40-4 HCAPLUS RN CN Terbium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2hydroxybenzaldehydato-O,O')- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** 175091-41-5 HCAPLUS RNNeodymium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2-CN hydroxy-5-methoxybenzaldehydato-O1,O2)- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** 175091-42-6 HCAPLUS RN CN Terbium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2-hydroxy-5-methoxybenzaldehydato-O1,O2)- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** 175091-25-5P 175091-26-6P 175091-29-9P IT 175091-34-6P 175091-35-7P RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of) 175091-25-5 HCAPLUS RNSamarium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2-CN hydroxybenzaldehydato-O,O')- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** RN 175091-26-6 HCAPLUS Gadolinium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2-CN hydroxybenzaldehydato-O,O')- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** 175091-29-9 HCAPLUS RN Erbium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2-CN hydroxybenzaldehydato-O,O')- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** RN 175091-34-6 HCAPLUS Samarium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2, N2', N2''](2-hydroxy-CN 5-methoxybenzaldehydato-O1,O2)- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** 175091-35-7 HCAPLUS RN Gadolinium, bis[hydrotris(1H-pyrazolato-N1)borato(1-)-N2,N2',N2''](2-CN hydroxy-5-methoxybenzaldehydato-O1,O2)- (9CI) (CA INDEX NAME) *** STRUCTURE DIAGRAM IS NOT AVAILABLE *** L16 ANSWER 15 OF 17 HCAPLUS COPYRIGHT 2002 ACS

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ΧU
      09/445050
                       Page 149
     1995:323915 HCAPLUS
ΑN
     122:150111
DN
     Synthesis of the new tripodal ligand tris-[3-(2'-pyridyl)pyrazol-1-
ΤI
     yl]hydroborate, and the crystal structure of its europium(III) complex
     Amoroso, Angelo J.; Thompson, Alexander M. Cargill; Jeffery, John C.;
ΑU
     Jones, Peter L.; McCleverty, Jon A.; Ward, Michael D.
     Sch. Chem., Univ. Bristol, Bristol, BS8 1TS, UK
CS
     J. Chem. Soc., Chem. Commun. (1994), (24), 2751-2
SO
     CODEN: JCCCAT; ISSN: 0022-4936
DT
     Journal
LA
     English
     78-7 (Inorganic Chemicals and Reactions)
CC
     Section cross-reference(s): 28, 75
     The new tripodal ligand tris-[3-(2'-pyridyl)pyrazol-1-yl]hydroborate (L-),
AB
     comprising three N,N,-bidentate chelating arms linked by the apical B
     atom, was synthesized; the crystal structure of [EuL(MeOH)2F][PF6] reveals
     the nine-coordinate metal lying within the hexadentate ligand cavity.
     crystal structure europium complex pyridylpyrazolylborate; borate
ST
     trispyridylpyrazolylhydro europium complex; pyrazolylborate pyridyl deriv
     europium complex
IT
    Crystal structure
       Luminescence
    Molecular structure
        (of europium complex with tris[(pyridyl)pyrazolyl]hydroborate)
                             302-01-2, Hydrazine, reactions
IT
     68-12-2, DMF, reactions
     2-Acetylpyridine
                        13762-51-1, Potassium tetrahydroborate(1-)
     RL: RCT (Reactant)
        (for prepn. of tris[(pyridyl)pyrazolyl]hydroborate and europium
        complex)
     75415-03-1P, 3-(2'-Pyridyl)pyrazole
                                           123367-25-9P
IT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
        (for prepn. of tris[(pyridyl)pyrazolyl]hydroborate and europium
        complex)
     161095-31-4P, Potassium tris[3-(2'-pyridyl)pyrazol-1-yl]hydroborate(1-)
ΙT
    RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and complexation with europium)
TΤ
     161095-30-3P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn., crystal structure and luminescence of)
     161095-30-3P
TΤ
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn., crystal structure and luminescence of)
     161095-30-3 HCAPLUS
RN
     Europium(1+), fluoro[hydrotris[2-(1H-pyrazol-3-yl-
CN
     .kappa.N1)pyridinato]borato(1-)]bis(methanol)-, hexafluorophosphate(1-)
     (9CI) (CA INDEX NAME)
     CM
          1
     CRN 161095-29-0
         C26 H27 B Eu F N9 O2
     CMF
     CCI
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CM 2

CRN 16919-18-9

CMF F6 P

CCI CCS

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DN 114:217117

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ΤI Luminescence studies of tris[dihydrobis(1pyrazolyl)borato]terbium(III) Reger, Daniel L.; Chou, Pi Tai; Studer, Shannon L.; Knox, Steven J.; ΑU Martinez, Marty L.; Brewer, William E. Dep. Chem., Univ. South Carolina, Columbia, SC, 29208, USA CS Inorg. Chem. (1991), 30(10), 2397-402 SO CODEN: INOCAJ; ISSN: 0020-1669 Journal DΨ LA English 73-5 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties) AΒ The luminescence spectra and dynamics of [H2B(pz)2]3Tb were studied at different temps., in the solid phase, and in various solvents. Anal. of the data for the cryst. sample based on the electron dipole selection rules reveals effective C3 symmetry. Thus, the luminescence spectra are those expected for the trigonal-prismatic arrangement of the nitrogen donor atoms, but are influenced by the three weak BH---Tb three-center bonds to each of the rectangular faces. Similar luminescence spectra were obsd. for [H2B(pz)2]3Tb in CH2Cl2 and toluene solns. In donor solvents, complexation of [H2B(pz)2]3Tb with the solvent mols. changes the lifetime and the spectral features of the luminescence, indicating a change in the coordination environment about [H2B(pz)2]3Tb. ST luminescence terbium dihydrobispyrazolylborato; hydrobispyrazolylborato terbium luminescence; pyrazolylborato hydro borato terbium luminescence; solvent effect terbium hydrobispyrazolylborato luminescence; THF solvent effect terbium hydrobispyrazolylborato luminescence; acetonitrile solvent effect terbium hydrobispyrazolylborato luminescence ΙT Luminescence (of tris(dihydrobispyrazolylborato)terbium, solvent effect on) IT Solvent effect (on luminescence, of trisdihydrobispyrazolylboratoterbium) IT 124755-02-8 RL: PRP (Properties) (luminescence of, crystal symmetry and solvent effect on coordination in relation to) 75-05-8, Acetonitrile, properties 109-99-9, properties IT RL: PRP (Properties) (solvent effect of, on fluorescence of trisdihydrobispyrazolylboratoter bium) 124755-02-8 IT RL: PRP (Properties) (luminescence of, crystal symmetry and solvent effect on

Terbium, tris[dihydrobis(1H-pyrazolato-N1)borato(1-)-H, N2, N2']-,

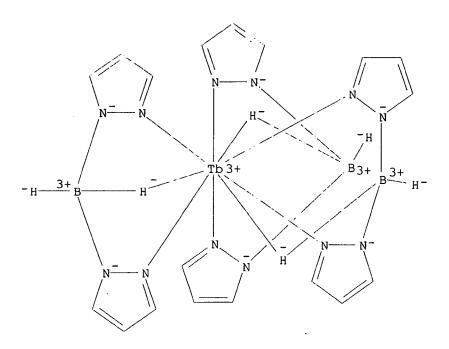
(TPS-9-1121'1'2'1''1''2'')- (9CI) (CA INDEX NAME)

coordination in relation to)

124755-02-8 HCAPLUS

RN

CN



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AN 1985:69597 HCAPLUS

DN 102:69597

TI Absorption and emission spectra of neodymium(III) and europium(III) complexes

AU Seminara, A.; Musumeci, A.

CS Ist. Dip. Chim. Chim. Ind., Univ. Catania, Catania, 95125, Italy

SO Inorg. Chim. Acta (1984), 95(6), 291-307 CODEN: ICHAA3; ISSN: 0020-1693

DT Journal

LA English

CC 73-4 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 78

AB The absorption and emission spectra of several complexes of Nd(III) and Eu(III) ions were examd. to obtain reliable information relating to coordination no., nature of bonding and symmetry around the lanthanide ion. Steric factors may force the polyhedron of coordination towards geometries less favorable by ligand-ligand repulsion. In general, no correlation was found between the low intensity of the hypersensitive transitions and high symmetry or low symmetry and high intensity. The results have pointed out the role of covalency in hypersensitivity.

ST UV visible europium neodymium complex; luminescence europium neodymium complex

IT Energy level

Luminescence

Ultraviolet and visible spectra

(of europium and neodymium complexes)

IT Bond

Coordination number

(of europium and neodymium in complexes)

IT Energy level transition

(hypersensitive, of europium and neodymium complexes)

IT 33461-92-6 47894-14-4 47894-18-8 47894-20-2 51187-22-5

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ΧU
     09/445050
                       Page 153
     51187-25-8
                  51221-90-0
                               53062-59-2
                                           55230-65-4
                                                        55230-66-5
     55237-82-6
                 55237-87-1
                               55238-02-3
                                           55238-04-5
                                                        59370-54-6
     67904-18-1
                  67904-27-2
                               67908-16-1
                                           67908-34-3
                                                         67908-35-4
     67908-36-5
                  67945-40-8
                               67965-57-5
                                           70948-77-5
                                                        70948-78-6
     89504-01-8
                  89554-45-0
                               89554-53-0 94369-53-6 94369-54-7
                  94369-56-9
                                           94369-58-1
                               94369-57-0
     94369-55-8
                                                        94369-82-1
                  94426-30-9
     94403-86-8
     RL: PRP (Properties)
        (absorption and emission spectrum of)
ΙT
     94369-53-6 94403-86-8
     RL: PRP (Properties)
        (absorption and emission spectrum of)
RN
     94369-53-6 HCAPLUS
     Europium, tris(hydrotris(1H-pyrazolato-.kappa.N1)borato(1-)-
CN
     .kappa.N2',.kappa.N2'',.kappa.N2'']-, (TPS-9-1111'1'1''1'')- (9CI) (CA
     INDEX NAME)
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RN 94403-86-8 HCAPLUS
CN Neodymium, tris[hydrotris(1H-pyrazolato-.kappa.N1)borato(1-).kappa.N2,.kappa.N2',.kappa.N2'']-, (TPS-9-1111'1'1''1'')- (9CI) (CFINDEX NAME)

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